

Mechanism of SK2 channel gating and its modulation by the bee toxin apamin and small molecules

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In this **important** manuscript, Cassell and colleagues set out on a mechanistic and pharmacological exploration of an engineered chimeric small conductance calcium-activated potassium channel 2 (SK2). They show **compelling** evidence that the SK2 channel possesses a unique extracellular structure that modulates the conductivity of the selectivity filter, and that this structure is the target for the SK2 inhibitor apamin. The interpretations are sound and the writing is clear, and the manuscript was strengthened during review by providing more detailed information for the electrophysiological experiments and the structural analyses attempted, in addition to relating dilation of the filter to mechanisms of inactivation in other potassium channels. This high-quality study will be of interest to membrane protein structural biologists, ion channel biophysicists, and chemical biologists, and will help to inform future drug development targeting SK channels.

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Abstract

Small-conductance calcium-activated potassium channel 2 (SK2) serves a variety of biological functions by coupling intracellular calcium dynamics with membrane potential. SK2 modulators are in development for the treatment of neurological and cardiovascular diseases, though the mechanisms of pharmacological modulation remain incompletely understood. We determined structures of an SK2-4 chimeric channel in Ca^{2+} -bound and Ca^{2+} -free conformations and in complex with the bee toxin apamin, a small molecule inhibitor, and a small molecule activator. The structures revealed that the S3-S4 linker forms a hydrophobic constriction at the extracellular opening of the pore. Apamin binds to this extracellular constriction and blocks the exit of potassium ions. Furthermore, we identified a structurally related SK2 inhibitor and activator that bind to the transmembrane domains. The compounds exert opposing effects on gating by differentially modulating the conformation of

the S6 helices. These results provide important mechanistic insights to facilitate the development of targeted SK2 channel therapeutics.

Introduction

Fundamental to a variety of biological processes, ion channels are an important class of therapeutical targets. Small conductance calcium-activated potassium (KCa2.x, or SK1, 2, and 3) channels are activated by increased intracellular Ca^{2+} to induce potassium efflux and regulate membrane potential^{1,2,3}. In this role, SK1, 2, and 3 mediate cellular excitability and have different but overlapping functions in many cell types including neurons, endothelial cells, and cardiomyocytes⁴. In particular, the SK2 channel regulates synaptic transmission and plasticity, learning and memory, and cardiac action potentials and thus has attracted attention as a potential target for the treatment of neurological and cardiovascular diseases^{2,5,6}. SK2 activators reduce cellular excitability and are potential therapeutics for alcohol dependence⁷, ataxia⁸, epilepsy⁹, and stroke¹⁰.

Conversely, SK2 inhibitors increase cellular excitability and have been proposed for the treatment of Alzheimer's disease¹¹ and atrial fibrillation¹².

The cryo-EM structure of the related SK4 (KCa3.1, IK) channel provided the first insights into the architecture and mechanism of Ca^{2+} -dependent gating for the SK channel family¹³. SK4 channels form non-domain swapped tetramers, with each subunit containing six transmembrane helices S1 to S6^{1,13}. S1-S4 form a voltage-sensor like domain where the S4 helices lack the positively charged residues necessary for voltage sensitivity. The S5 and S6 helices form the potassium pore. Within the potassium pore lies the selectivity filter, a structure unique to potassium channels that is required for rapid and selective conductance of K^+ ions¹⁴. Following the S6 helices there are two intracellular helices (HA and HB) that form the binding site for the Ca^{2+} -binding protein calmodulin (CaM), which acts as the Ca^{2+} sensor to gate SK channels¹⁵. The CaM C-lobe is constitutively bound to the HA and HB helices, and upon an increase in intracellular Ca^{2+} the CaM N-lobe binds to a unique S4-S5 linker, inducing a conformational change in the S6 helices to open the potassium pore and activate the channel¹³.

SK2 activators described to date bind at the interface of the CaM N-lobe and S4-S5 linker and function by stabilizing this interaction^{13,16,17}. On the other hand, known SK2 inhibitors target the extracellular and/or transmembrane regions and were proposed to function by either direct pore block or negative gating modulation¹⁶. The binding site for the bee venom toxin apamin, a cyclic 18 residue peptide inhibitor, has been mapped to the extracellular loop regions of SK2¹⁸⁻²⁰. Apamin is of historical importance as it was used to elucidate the physiological role of SK2 and apamin inhibition of SK2 increases neuronal excitability and improves learning and memory^{3,21}. Apamin inhibits SK1, 2, and 3 but not 4, and is most potent against SK2 with an IC_{50} of ~ 70 pM^{1,22}. Functional mutagenesis of apamin identified two arginine residues that are essential for inhibition²³. Mutagenesis experiments on the SK2 channel indicated that residues in the extracellular loops between S3 and S4 (S3-S4 linker) and between S5 and S6 are important for apamin binding and inhibition¹⁸⁻²⁰. Since the S3-S4 linker is predicted to be distant to the pore an allosteric mechanism of apamin inhibition rather than a direct pore block has been suggested²⁰. Attempts to recapitulate the potency and selectivity of apamin with small molecules resulted in the development of a class of inhibitors, such as UCL1684, that are predicted to have an overlapping binding site with apamin²⁴⁻²⁶. These small molecules generally carry two positive charges, which may mimic the two arginine residues in apamin that are essential for inhibition²³. Further characterization of the molecular mechanisms of inhibition by apamin and the small molecule pore blockers is required to understand how the interaction between the S3-S4 linker and the essential arginine residues/positive charges block ion conduction in SK2.

Another class of small molecule SK2 inhibitors act as negative gating modulators by shifting the Ca^{2+} dependence of activation to higher Ca^{2+} concentrations^{27,28}. One such inhibitor, AP31969, is currently in clinical trials for the treatment of arrhythmia²⁹.

Mutagenesis experiments with the structurally-related inhibitor AP14145 suggest that these compounds bind within the pore directly below the selectivity filter²⁸. Similar potencies on SK1, 2, and 3 channels have been reported most likely due to the homology of pore lining residues in the SK family. However, selective SK1 inhibitors were developed that take advantage of a unique residue, Ser293, on S5³⁰. Interestingly, only small modifications to this family of inhibitors are sufficient to switch the activity profile from inhibition to activation of SK1. However, it remains unclear how compounds that bind the transmembrane regions of SK channels affect Ca^{2+} -dependent gating, which is driven by the interaction between CaM and the intracellular domains.

Despite SK2 being a prominent therapeutic target for both neurological and cardiovascular diseases, no structure of human SK2 has been reported to date. Although, the structure of rat SK2 was reported while this manuscript was in preparation³¹. To enable high-resolution cryo-EM studies of human SK2, we designed a chimera (SK2-4) that contains the transmembrane and extracellular domains of human SK2 and intracellular domains of human SK4. The structures of the SK2-4/CaM complexes in the Ca^{2+} -bound and Ca^{2+} -free conformations demonstrate that SK2 and SK4 adopt similar overall architectures and share a similar mechanism for Ca^{2+} -dependent gating. However, unlike SK4, we observed a structured S3-S4 linker that induces a conformational change in the selectivity filter and forms a hydrophobic constriction at the extracellular opening of the SK2 pore. Apamin binds to the extracellular constriction formed by the S3-S4 linker to block potassium efflux. In addition, high throughput screening and medicinal chemistry optimization efforts yielded a new class of potent SK2 inhibitors that bind to a novel pocket formed by the S5, S6, and pore helices and induce closure of the S6 helices. Structure-guided design efforts enabled switching the activity profile towards activation while retaining the same binding mode. The detailed understanding of two distinct mechanisms of SK2 channel inhibition, extracellular pore block and negative gating modulation, and a new mechanism for channel activation presented here should facilitate the rational design of potent and selective SK2 modulators.

Results

Characterization of the SK2-4 chimeric channel

We attempted to solve the structure of wild-type (WT) human SK2 (Fig S1) expressed and purified from mammalian cells in the presence of CaM and Ca^{2+} . However, 2D class averages showed disorder in the intracellular region (Fig S2A). Subsequent 3D reconstruction generated a low-resolution, anisotropic map for the transmembrane helices and no interpretable density for the intracellular domains.

The SK4 channel cryo-EM structure suggested that the intracellular regions of SK4 may be more rigid than those of SK2¹³. We hypothesized that the SK4 intracellular domains would remain ordered when fused to the SK2 TM domains. A chimeric construct was created with the goal of preserving all transmembrane and extracellular regions of human SK2 while replacing the N- and C-terminal intracellular regions with those of human SK4 (Fig 1A, S1, and methods for details). Within the intracellular regions, only the N-terminal portion of the HA helix (residues 401-412 in SK2) that was predicted to interact with the S4-S5 linker retained the SK2 sequence¹³.

To confirm that the SK2-SK4 chimeric channel (SK2-4) behaves similarly to the WT SK2 channel, we compared the function and pharmacology of SK2-4 overexpressed in CHO cells to WT human SK2 and SK4 using an automated electrophysiology instrument, the Qube (Sophion). SK currents were recorded in the “whole-cell” configuration under voltage clamp where a saturating

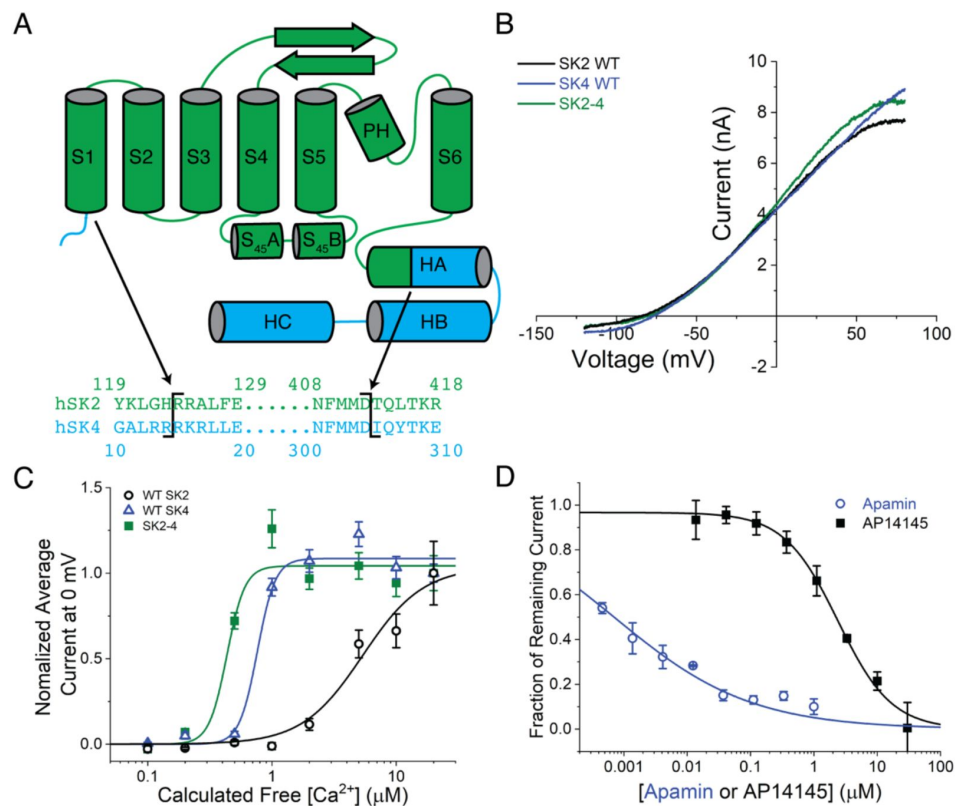


Figure 1

Design and characterization of SK2-4 chimera.

A) Cartoon representation of the SK2-4 chimera. Residue numbers in the SK2 regions of the structure correspond to the human SK2 sequence described by Desai *et al.*³² and the SK4 regions of the structure are numbered sequentially from the SK2 regions. The sequence alignment of human SK2 (green) and human SK4 (cyan) indicate the chimera boundaries (black lines). B) Representative current traces recorded with the Qube instrument in response to a ramp voltage protocol from -120 mV to 80 mV show that reversal potentials of WT SK2 (black), SK4 (blue), and SK2-4 (green) currents are all around -85 mV demonstrating similar potassium selectivity. Shown SK current traces are isolated by subtracting leak current determined under saturating concentration of inhibitor (UCL1684 or TRAM-34). Intracellular solution contains 2 μM Ca^{2+} based on calculation using Maxchelator (see methods) C) SK2-4 (green squares) is activated by increasing intracellular Ca^{2+} like WT SK2 (black circles) and SK4 channels (blue triangles). For each channel type, average SK current amplitudes from different cell populations each exposed to a different intracellular free Ca^{2+} concentration (calculated) ranging from 0.1 to 20 μM were normalized to average value at 20 μM Ca^{2+} . Apparent EC_{50} s for SK2, SK4 and SK2-4 are 5.4 μM , 0.8 μM and 0.4 μM , respectively. Note that actual intracellular free $[\text{Ca}^{2+}]$ might differ from intracellular buffer due to limit of diffusion in whole-cell configuration. Data points reflect mean $\pm$ SEM ($n=24$) D) SK2-4 is inhibited by apamin (blue circles, $\text{IC}_{50}=0.7$ nM), which targets the extracellular domains of SK2, and AP14145 (black squares, $\text{IC}_{50}=2.4$ μM), which targets the transmembrane domains of SK2. Data points reflect mean $\pm$ SD ($n=4$). Change in SK2-4 current amplitude for different cell populations each exposed to a different extracellular compound concentration was normalized to DMSO control to construct dose responses.

concentration of a specific SK2 inhibitor was applied at the end of experiment to isolate SK current. Electrophysiological measurements demonstrated that SK2-4 current has a similar reversal potential as WT SK2 and SK4 currents (**Fig 1B**), which is near the expected value for a K^+ -selective channel based on the K^+ concentration gradient. To evaluate Ca^{2+} dependent activation, we measured average SK current amplitude from different cell populations each exposed to a different intracellular free Ca^{2+} concentration ranging from 0.1 to 20 μM (i.e. each cell was only tested with an individual free intracellular Ca^{2+} concentration). Like WT SK2 and SK4, the activation of SK2-4 is dependent on intracellular Ca^{2+} concentration (**Fig 1C**).

Despite the potential limit in diffusion with the “whole-cell” configuration which precludes the ability to fully control the intracellular free Ca^{2+} concentration, our results clearly showed that the Ca^{2+} sensitivity of SK2-4 is higher than that of SK2 but similar to SK4, consistent with the CaM binding domain of the chimera stemming from SK4. In addition, SK2-4 activity was inhibited by known SK2 inhibitors apamin and AP14145 with an IC_{50} of 0.7 nM and 2.4 μM , respectively (**Fig 1D**). Of note, neither of these inhibitors is active against the WT SK4 channel, indicating that key regions of the SK2 native structure are preserved in the chimera. We also tested dozens of other available SK2 inhibitors of diverse scaffolds and observed a close correlation between potency on WT SK2 and the SK2-4 chimera (data not shown). Given the high degree of agreement between our characterization and previously reported data, we are confident that SK2-4 recapitulates the native activity and transmembrane architecture of WT SK2.

Structure of SK2-4/CaM in Ca^{2+} -bound and Ca^{2+} -free states

SK2-4 was co-expressed with CaM and purified from mammalian cells in the presence of Ca^{2+} . The cryo-EM structure was determined to a final resolution of 3.1 Å with most regions of the structure well-defined by density apart from the C-terminal HC helical bundle (**Fig 2A**, S2, S3A, Table 1). Residue numbers in the SK2 regions of the structure correspond to the human SK2 sequence described by Desai *et al.* and the SK4 regions of the structure are numbered sequentially from the SK2 regions (**Fig 1A**).

SK2-4/CaM forms a non-domain swapped tetramer with four CaM molecules bound, similar to the previously reported SK4/CaM structure. In the presence of Ca^{2+} , the C-lobe of each CaM is bound to the HA and HB helices and the N-lobe is bound to a short helix in the S4-S5 linker, S₄₅A, on the neighboring subunit (Fig S2G,H). The measured SK2-4 pore radius (3.8 Å) indicates that the S6 helices are in an open conformation in the presence of Ca^{2+} (**Fig 2C,D**).

We also purified SK2-4/CaM in the absence of Ca^{2+} and determined the cryo-EM structure at 3.4 Å (**Fig 2B**, S2, S3B, Table 1). In the structure, there is clear density for the S1-S6 regions but as observed in the closed conformation of SK4, the CaM density is weak and the N-lobes of CaM are dissociated from the S₄₅A helices. Furthermore, the S₄₅B and S6 helices collapsed around the pore axis and the pore is closed with a radius of 1 Å at the intracellular gate at Val390 (**Fig 2C,D**). The structures of SK2-4/CaM in the Ca^{2+} -bound and Ca^{2+} -free conformations confirm that the chimeric channel is gated by Ca^{2+} and that the mechanism of channel gating is similar to that of SK4.

S3-S4 linker structure

Although SK2-4 and SK4 share similar overall architecture, there are notable differences in the SK2 portion spanning the S1-S6 transmembrane region. The extracellular S3-S4 linker is not visible in the structure of SK4, indicating flexibility and lack of defined structure. In contrast, the S3-S4 linker of SK2-4 is well-defined by density in both the Ca^{2+} -bound and Ca^{2+} -free structures and forms a two-stranded anti-parallel β -turn (residues Gly231-Asp253) that extends over the S5 and S6 helices (**Fig 3**, S3A,B). The β -turn interacts with residues Tyr335 and His336 at the C-terminus of the S5 helix (**Fig 3B**). His336 forms an edge-to-face interaction with Trp237 and a hydrogen bond with Ser248, whereas Tyr335 forms a hydrogen bond with Asp253. The β -turn features an eight

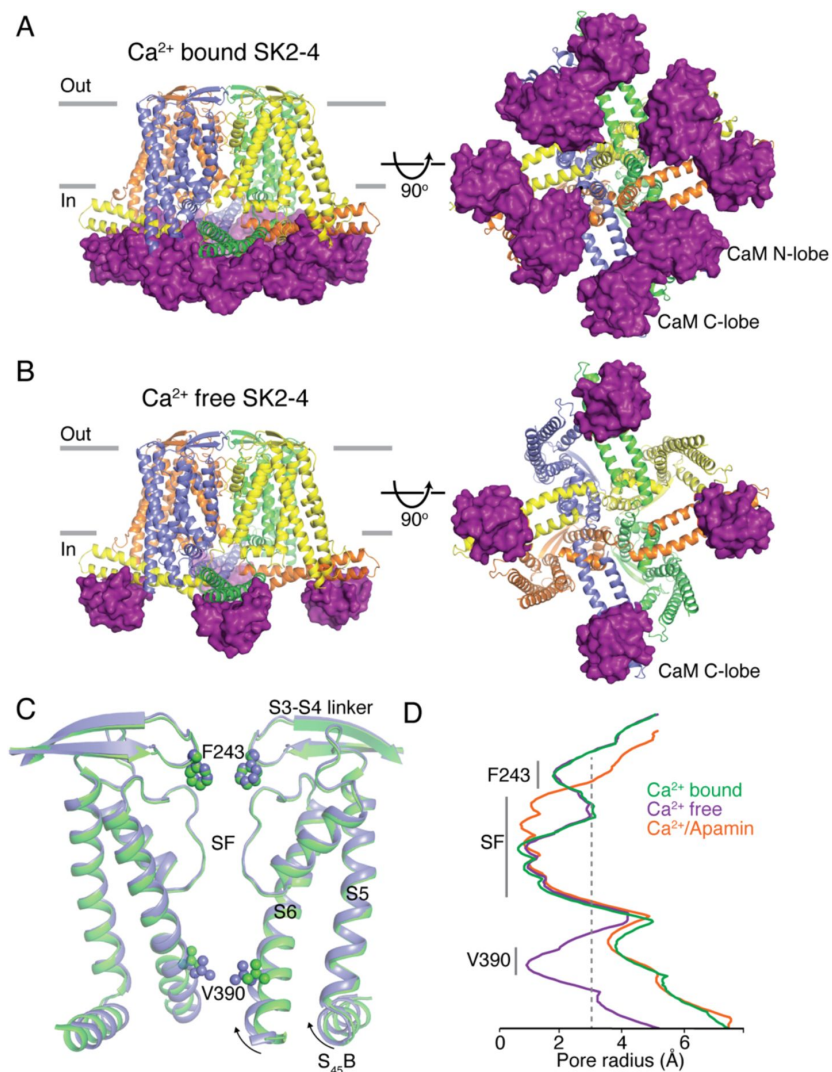


Figure 2

SK2-4 chimera architecture.

Structures of Ca^{2+} -bound (A) and Ca^{2+} -free (B) SK2-4. SK2-4 is shown as a cartoon with each subunit of the tetramer in a different color. The grey lines (left) indicate membrane boundaries. In Ca^{2+} -bound SK2-4 (A) both the N- and C-lobes of CaM (purple surface) are associated with the intracellular domains. In Ca^{2+} -free SK2-4 (B) only the CaM C-lobe is bound (purple surface) and the N-lobe is dissociated (not shown). C) Overlay of the K^+ pore from Ca^{2+} -bound (green) and Ca^{2+} -free (lavender) SK2-4 structures. Same view as left panels of (A) and (B) but only 2 subunits are shown. D) Pore radii of Ca^{2+} -bound (green), Ca^{2+} -free (lavender), and apamin-bound (orange) SK2-4. The location of the intracellular gate (Val390), selectivity filter (SF), and extracellular constriction (Phe243) are indicated. Grey dashed line indicates the radius of hydrated K^+ (3 Å).

amino acid loop (residues 240-247) that extends into the potassium pore (**Fig 3C**, 4A, S4A,B). Arg240, Phe243, and Tyr245 within this loop form direct interactions with residues at the C-terminus of the selectivity filter. Arg240 and Tyr245 form a salt bridge and hydrogen bond with the side chain and backbone of Asp363 from a neighboring subunit, respectively. Phe243 extends into the ion conduction path and is within C-H/O bonding distance to the backbone carbonyl of Gly362 (**Fig 3C**). In this position, the four Phe243 residues form edge-to-face interactions that create a hydrophobic constriction at the extracellular opening of the pore with a radius of 1.8 Å, which is expected to prohibit efflux of hydrated K^+ ions measuring 6 Å in diameter (**Fig 2D**, 3A, 4A). Therefore, even though the intracellular gate at Val390 is open in the SK2-4/CaM Ca^{2+} -bound structure, the observed conformation is expected to be non-conductive. For conduction to occur, a structural rearrangement of the S3-S4 linker that increases the diameter of the extracellular constriction would be required.

Focused 3D classification of the S3-S4 linker was unsuccessful in identifying particle subsets with a dilated extracellular constriction suggesting that either none or a very small percentage of Ca^{2+} -bound SK2-4 is in a conductive state. This along with the similar conformation of the S3-S4 linker in the Ca^{2+} -bound and Ca^{2+} -free states of SK2-4 suggest that Ca^{2+} -dependent intracellular gate dynamics are not coupled to the conformation of the S3-S4 linker. Other yet to be identified physiological factors are likely required to dilate the extracellular constriction.

All residues participating in interactions between the S3-S4 linker and the pore residues are fully conserved in SK1, 2, and 3 (Fig S1), suggesting that S3-S4 linker conformation is an important structural feature of the SK channels. In addition, the S3-S4 linker conformation is important for apamin inhibition as mutation of Tyr245 and His336, which are involved in hydrogen-bonding interactions between the S3-S4 linker and pore residues, reduce apamin potency^{19,20}.

Selectivity filter conformation

The selectivity filter is a crucial component of all selective K^+ channels, including SK4. It contains the highly conserved (T/S)XG(Y/F)G motif, where the T/S hydroxyl and the backbone carbonyls of subsequent residues form four consecutive K^+ coordination sites that perfectly mimic the hydration shell of a K^+ ion (**Fig 4B,C**)^{14,33}. Extensive structural and mechanistic studies demonstrated that this selectivity filter structure is required for the fast and selective conduction of K^+ ions³⁴⁻³⁸. Reducing the number of K^+ coordination sites in the selectivity filter produces channels that are either not K^+ selective, like HCN or NaK, or not conductive, like voltage gated potassium channels (K_v) that have undergone C-type inactivation (**Fig 4E,F,G,H**)^{36,39-41}.

In the structures of SK2-4/CaM in both the Ca^{2+} -bound and Ca^{2+} -free conformations, Tyr361, located in the center of the selectivity filter, is rotated approximately 180° when compared with the homologous Tyr253 in SK4 (**Fig 4A,C,D**, S4A,B)¹³. In SK4, Tyr253 hydrogen bonds with a pore helix threonine (Thr247) and the homologous interaction is not observed in SK2-4 due to the rotation of Tyr361. In addition, the backbone carbonyls of Gly360 and Tyr361, which typically form the extracellular K^+ coordination sites one and two, are rotated away from the center of the selectivity filter and the ion conduction path. As a result, only the two intracellular K^+ coordination sites, three and four, are intact and occupied by K^+ . The selectivity filter of SK2-4 resembles that of HCN in both the position of Tyr361 and the number of ion coordination sites (**Fig 4E,F,G,H**)^{36,39-41}.

SK2 and SK4 selectivity filters were predicted to adopt similar conformations based on sequence conservation in the selectivity filter and surrounding residues (Fig S1).

Therefore, the conformation of the selectivity filter observed in the SK2-4 structure is likely due to the S3-S4 linker, which is disordered in SK4. As discussed above, Asp363 at the C-terminus of the selectivity filter is directed towards the extracellular S3-S4 linker and interacts with Arg240 and Tyr245 (**Fig 3C**, 4A, S4A,B). In SK4, the homologous aspartate (Asp255) is directed towards the

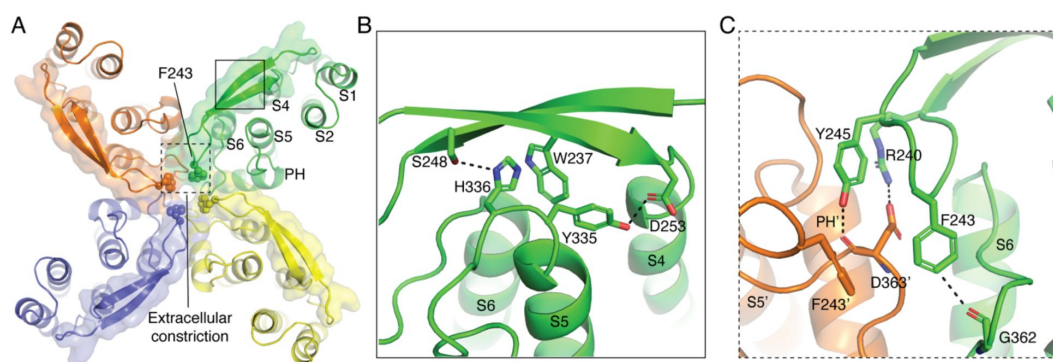


Figure 3

S3-S4 loop architecture.

A) Extracellular view of Ca²⁺-bound SK2-4 with each subunit of the tetramer in a different color. The S3-S4 linker (surface and cartoon) extends over the S5 and S6 helices. Phe243 residues (spheres) form an extracellular constriction with a radius of 1.8 Å. Boxes indicate location of the interactions shown in (B) and (C). B) Interactions between the S3-S4 linker and the C-terminus of S5. His336 forms an edge-to-face interaction with Trp237 and hydrogen bond with Ser248 (dashed lines). Tyr335 forms a hydrogen bond with D253 (dashed lines). C) Interactions between the S3-S4 linker and the C-terminus of the selectivity filter. Arg240 and Tyr245 (green sticks) from the S3-S4 linker form a salt bridge and hydrogen bond (dashed lines) with the side chain and backbone carbonyl of Asp363 from the neighboring subunit (orange sticks), respectively. Phe243 (green sticks) forms an edge-to-face interaction with the neighboring Phe243 (orange sticks) and is in position to form a C-H/O interaction (dashed line) with Gly262 (green sticks) from the same subunit.

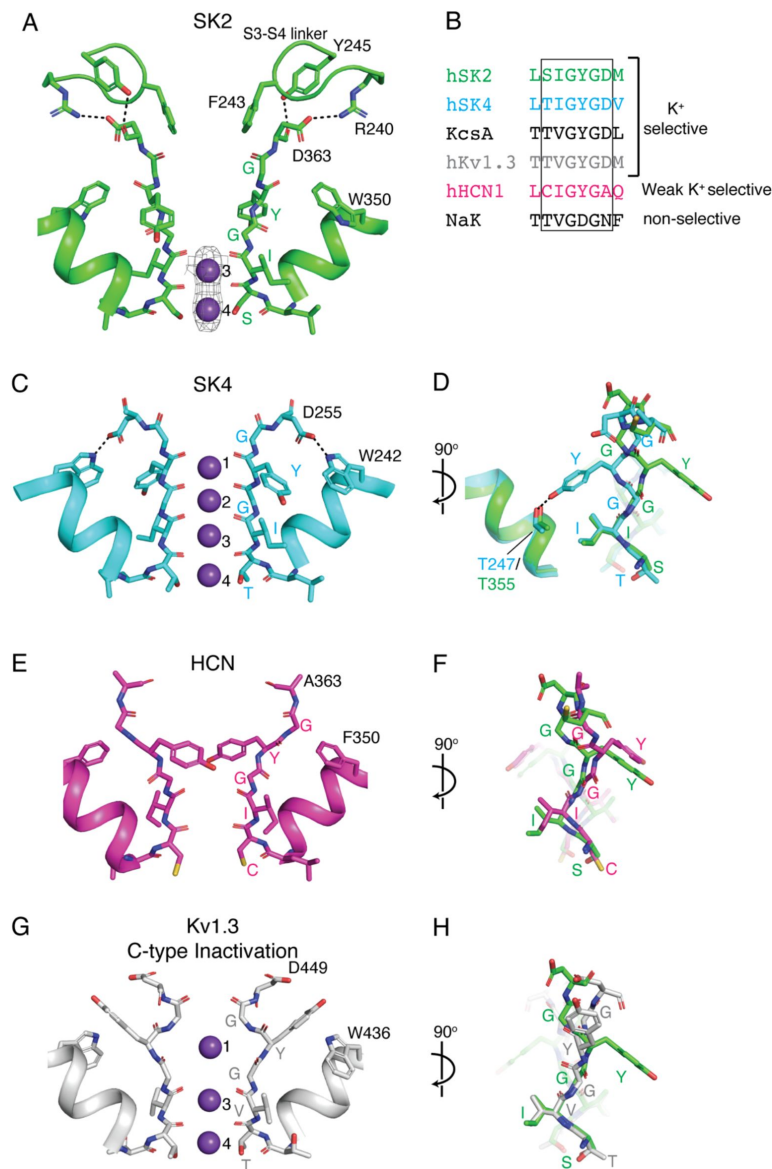


Figure 4

Selectivity filter conformation of SK2-4.

A) Ca²⁺-bound SK2-4 selectivity filter structure (green). Arg240 and Tyr245 from the S3-S4 linker form a salt bridge and hydrogen bond (dashed lines) with Asp363 and the selectivity filter adopts a conformation with two K⁺ coordination sites (purple spheres, density shown at 5 σ). B) Sequence alignment of selectivity filter from K⁺ selective channels (hSK2, hSK4, KcsA, hKv1.3) and non-selective cation channels (hHCN and NaK) C) Structure of hSK4 selectivity filter (cyan, PDB: 6CNN). In SK4 there is a conserved hydrogen bond between the selectivity filter Asp255 and pore helix Trp242 (dashed lines), Tyr253 hydrogen bonds with Thr247, and the selectivity filter conformation creates four occupied K⁺ coordination sites (purple spheres). D) Overlay of the SK2-4 (green) and the SK4 (cyan) selectivity filter with a 90° rotation from (C). E) Structure of HCN selectivity filter (magenta, PDB: 5U6O). In HCN there is no interaction between the Ala363 at the C-terminus of the selectivity filter and the pore helix Phe350, Tyr361 is in a similar conformation to Ca²⁺-bound SK2-4, and the selectivity filter adopts a conformation with two ion coordination sites (the authors did not model the ions but cryoEM density indicates two coordination sites). F) Overlay of the SK2-4 (green) and the HCN (magenta) selectivity filter with a 90° rotation from (E). G) Structure of Kv1.3 selectivity filter in a C-type inactivated conformation (magenta, PDB: 7WF3). In this conformation there is no interaction between Asp449 at the C-terminus of the selectivity filter and the pore helix Trp436, Tyr447 is directed extracellularly, and the selectivity filter adopts a conformation with three occupied K⁺ coordination sites (purple spheres). H) Overlay of the SK2-4 (green) and the Kv1.3 (grey) selectivity filter with a 90° rotation from (G).

pore helix and forms a hydrogen bond with a conserved tryptophan (Trp242) (**Fig 4C**)¹³. The homologous pore helix tryptophan in SK2, Trp350, adopts a different rotamer and is unable to form a hydrogen bond with Asp363 (**Fig 4A**, S3A,B). The hydrogen-bonding interaction between a pore helix residue (usually Trp or Tyr) and the residue at the C-terminus of the selectivity filter (usually Asp or Asn) is a conserved feature of K⁺ selective channels³⁸. Furthermore, inserting this interaction into non-selective channels produces K⁺ selective channels with four K⁺ coordination sites, while deletion of this interaction from K⁺ selective channels produces non-selective channels with two K⁺ coordination sites^{37,38}. Therefore, we propose that by interacting with Asp363 and thereby preventing its interaction with the pore helix Trp350, the S3-S4 linker induces the selectivity filter conformation observed in SK2-4 with two K⁺ coordination sites. We predict that the SK1 and 3 selectivity filters could adopt similar conformations with two K⁺ coordination sites because the S3-S4 linker residues that interact with Asp363 are conserved among these channels (Fig S1).

Except for one report indicating significant sodium permeability in rat SK2⁴², studies show that SK2 is K⁺ selective^{1,32,43,44}. The reversal potential of SK2 and SK2-4 presented here confirms K⁺ selectivity (**Fig 1B**). This contrasts with the SK2-4 structures as selectivity filters with two K⁺ coordination sites are predicted to be non-selective^{36–39}. However, the Ca²⁺-bound SK2-4/CaM structure is in a non-conductive conformation, despite the open intracellular Val390 gate, due to Phe243 of the S3-S4 linker forming a constriction at the extracellular opening that prevents K⁺ efflux. Thus, the selectivity filter conformation should not affect the K⁺ selectivity observed in the experimental data^{1,32,43,44}.

Mechanism of apamin inhibition

Extensive mutational studies suggest that the SK2 S3-S4 linker is essential for apamin binding and inhibition, although the nature of their interaction remains unknown^{18,19}. To understand how apamin interacts with the unique conformation of the S3-S4 linker and how that interaction inhibits K⁺ conduction, we incubated Ca²⁺-bound SK2-4/CaM with excess apamin prior to cryo-EM grid preparation. 3D reconstruction in C4 symmetry produced a map with additional density at the extracellular opening of the K⁺ pore consistent with the size of a single apamin molecule bound per tetramer. However, the C4 symmetric density corresponds to asymmetric apamin bound to four distinct but structurally equivalent orientations, which prevented apamin fitting. To more accurately align particles and improve the apamin density we used a combination of focused classification and local refinements to generate a reconstruction in C1 symmetry to a final resolution of 3.2 Å (Fig S3C, S5, Table 1). Elongated density in the C1 map at the extracellular opening of the pore allowed for accurate placement of the C-terminal helix (residues 6-17) from an NMR structure of apamin (**Fig 5A**)²². Two well-defined tubular densities at the center of the helix were confidently modeled with the side chains of Arg13 and Arg14.

Overall, the apamin-bound SK2-4/CaM structure resembles Ca²⁺-bound SK2-4. The N-terminal lobe of CaM engages with the S₄₅A helix, the S5 and S6 helices adopt a similar conformation, and the intracellular gate Val390 is open with a radius of 3.5 Å (**Fig 2D**).

The most significant conformational change is in the position of the S3-S4 linker, which shifts ~2 Å away from the pore axis to accommodate apamin binding (**Fig 5C**). The C-terminal helix of apamin binds at the extracellular opening of the pore and interacts with all four S3-S4 linkers of the tetramer (**Fig 5A**). Essential residues Arg13 and Arg14 insert into the Phe243 constriction to form cation- π interactions (**Fig 5A,C**)²³. In this position, apamin blocks the exit of K⁺ ions from the extracellular side of the pore to inhibit conduction.

To probe the interaction between SK2 Phe243 and apamin Arg13 and Arg14 we determined the apamin sensitivity of a Phe243Ala mutant and found it to be insensitive to apamin inhibition up to 3 μ M (**Fig 5B**), demonstrating the critical role of this interaction for apamin binding and inhibition. Prior mutagenesis of both the S3-S4 linker and the outer pore regions further support

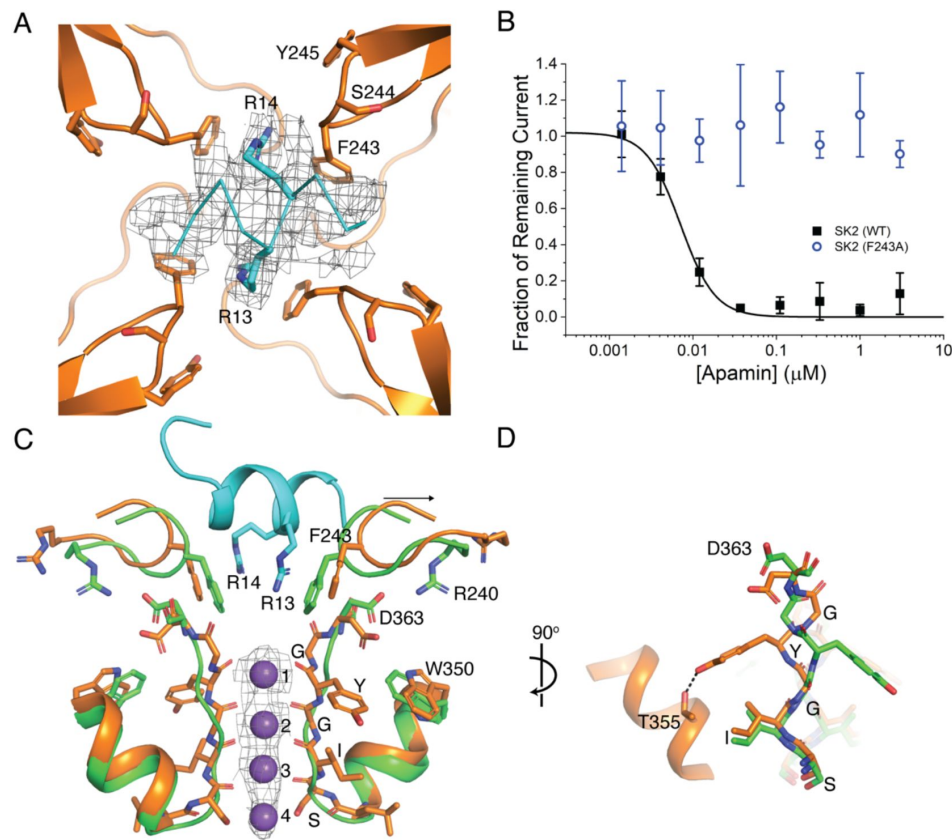


Figure 5

Mechanism of apamin inhibition.

A) Extracellular view of apamin binding site. Apamin (cyan ribbon, density shown at 5 σ) binds to the S3-S4 extracellular gate. Apamin Arg13 and Arg14 (cyan) and the S3-S4 linker residues (orange) that surround the apamin binding site are shown as sticks. B) Apamin sensitivity of WT SK2 (black squares) and SK2 F243A (blue circles). Change in SK current amplitude for different cell populations each exposed to a different extracellular apamin concentration was normalized to DMSO control to construct dose responses. Data points represent mean $\pm$ SD (n=6). The F243A mutant is insensitive to apamin up to 3 μ M. C) Overlay of Ca^{2+} -bound SK2-4 selectivity filter (green) and apamin-bound SK2-4 selectivity filter (orange, selectivity filter shown as sticks and S3-S4 linker and pore helix shown as a cartoon). Upon apamin (cyan) binding the S3-S4 linker retracts from the pore axis (black arrow). Asp363 reorients into hydrogen bonding position with a rotated Trp350 side chain, Tyr361 in the selectivity filter is rotated 180° to form a hydrogen bond with Thr355 in the pore helix, and the selectivity filter adopts a conformation with four K^+ coordination sites (purple spheres, density shown at 5 σ). D) 90° degree rotation of (C).

the observed apamin binding site and proposed mechanism of apamin inhibition. For example, mutation of Ser244 and Tyr245 in the S3-S4 linker reduce the potency of apamin by 10-fold and 5-fold, respectively^{18,19}. Ser244 is within 5 Å of the apamin binding site and bulky mutations may clash with apamin (**Fig 5A**). Tyr245 hydrogen bonds with Asp363 likely stabilizing the S3-S4 linker conformation for apamin binding (**Fig 3C**). In the outer pore, mutation of His336 decreases sensitivity of SK2 to apamin²⁰. As discussed above, this residue also stabilizes the S3-S4 linker conformation (**Fig 3B**). Notably, the previous mutagenesis experiments decreased but did not fully abolish the potency of apamin as we observed for the Phe243Ala mutation underscoring that the interaction between apamin and Phe243 is essential.

Fortuitously, the apamin-bound structure provides some insight into S3-S4 linker dynamics and the potential transition from a closed to open extracellular constriction. Insertion of apamin Arg13 and Arg14 dilates the Phe243 constriction from 1.8 Å to 3 Å (**Fig 2D** and **5C**). Such an S3-S4 linker conformation in the absence of apamin, would permit K⁺ ion conduction, providing an example of the type of S3-S4 movement required to expand the extracellular constriction. Compared with the apamin-free Ca²⁺-bound conformation (**Fig 4A**), the movement of the S3-S4 linker away from the pore axis upon apamin binding is accompanied by multiple conformational changes in the selectivity filter to adopt the canonical K⁺-selective conformation with four K⁺ coordination sites: Arg240 withdraws and frees Asp363 to form a weak hydrogen bond with a rotated Trp350 side chain while Tyr361 in the selectivity filter is rotated 180° and forms a hydrogen bond with Thr355 in the pore helix (**Fig 5C,D** and **S4C**). Toxin induced selectivity filter conformational change has also been reported for K_v1.3 with the sea anemone toxin ShK. However, unlike apamin binding to SK2-4, ShK binds directly to the K_v1.3 selectivity filter to convert a C-type inactivated conformation to a canonical K⁺ selective structure with four coordination sites^{40,41}. The change in selectivity filter conformation in apamin-bound SK2-4 seems to be driven instead by the weakening of interactions between the selectivity filter and the S3-S4 linker. Based on the structure of SK2-4 bound to apamin, we hypothesize that in the physiological conductive/dilated state of the extracellular constriction, the S3-S4 linker and Phe243 will withdraw from the pore axis and the selectivity filter will return to the canonical conformation with 4 K⁺ coordination sites to maintain the K⁺ selectivity demonstrated for SK2^{1,32,43,44}.

Identification and functional characterization of novel SK2 modulators

To identify new modulators of SK2 we developed a high throughput patch clamp assay using CHO cells stably expressing WT human SK2 channels on the Qube instrument (Sophion) (see methods section for details). Inhibition or activation of SK channels by compounds were characterized in the presence of 2 μM intracellular free Ca²⁺. Small molecule library screening and subsequent rounds of medicinal chemistry optimization produced compound 1 as a potent inhibitor of SK2 with an IC₅₀ of 69 nM (**Fig 6A,B**).

Selectivity profiling demonstrated that compound 1 has a 10-fold reduced potency for SK4 (IC₅₀ of 660 nM) but is selective over other ion channels tested including hERG, Nav1.5, KCNQ1, and Cav1.2 (**Fig 6B**, **S6A**). Compound 1 inhibits the SK2-4 chimera with similar potency as WT SK2 (**Fig 6B**), suggesting binding at the transmembrane or extracellular domains. Furthermore, we used differential scanning fluorimetry (DSF) to confirm binding of compound 1 to SK2-4. Addition of compound 1 resulted in a strong thermal stabilization of 10.8 °C, indicative of target engagement (**Fig 6C**).

To characterize the mechanism of inhibition, we determined the co-structure of compound 1 bound to SK2-4/CaM in the presence of Ca²⁺ at 3.3 Å resolution (**Fig S3D, S5, Table 1**). Four molecules of compound 1, which is clearly defined by cryo-EM density, were bound per SK2-4 tetramer at the interface of the S5, pore helix, and S6 of each subunit (**Fig 6D,E**). The

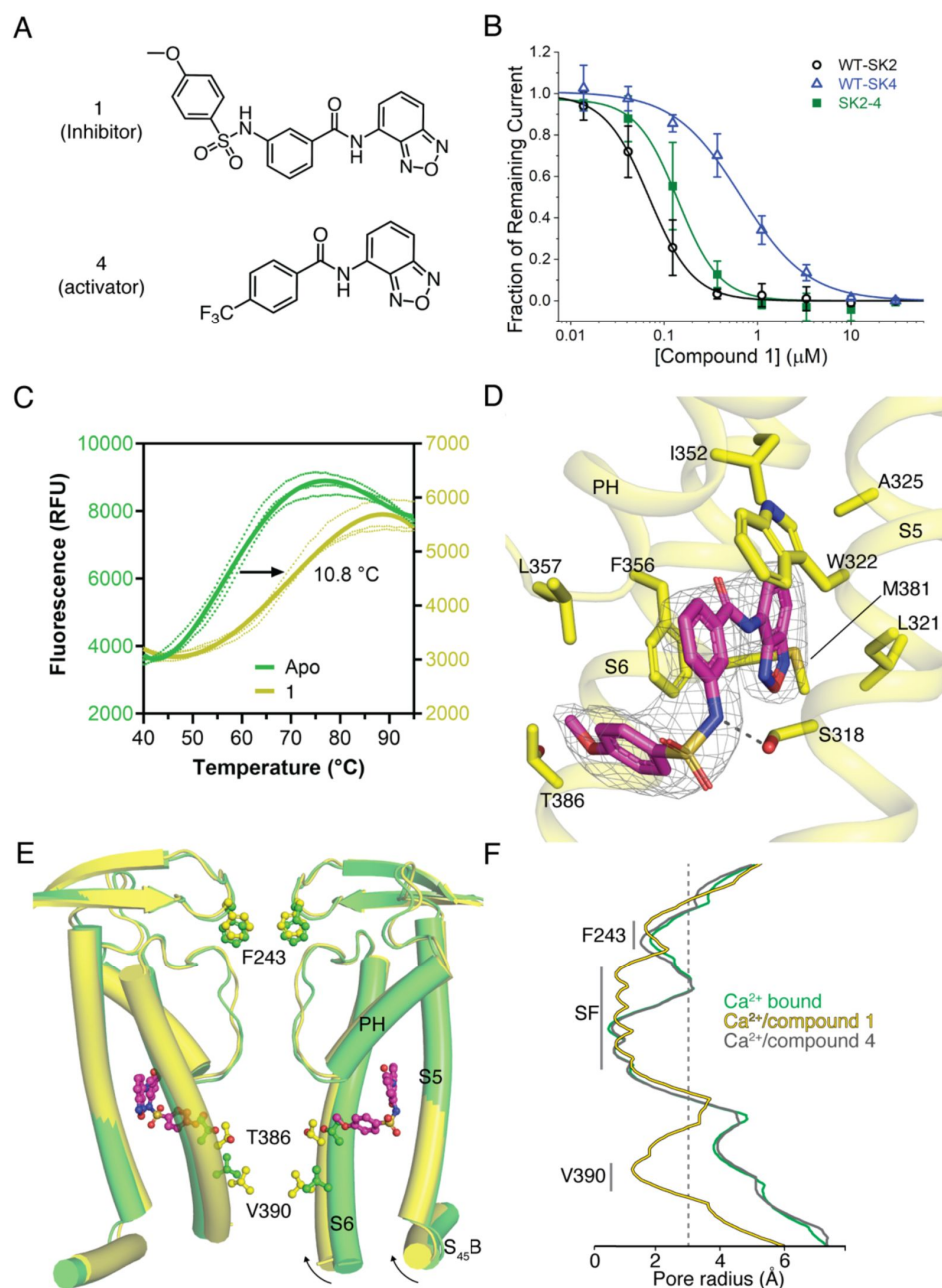


Figure 6

Mechanism of compound 1 inhibition.

A) Structure of compound 1 (inhibitor) and compound 4 (activator). B) Potency of compound 1 on SK2 (black circles, IC_{50} =69 nM), SK2-4 (green squares, IC_{50} =140 nM), and SK4 (blue triangles, IC_{50} =660 nM). Change in SK current amplitude for different cell populations each exposed to a different extracellular compound 1 concentration was normalized to DMSO control to construct dose responses. Data points represent mean $\pm$ SD (n =4). C) Melting curves of Ca^{2+} -bound SK2-4 in the absence (green, T_m = 58.7 °C) and presence (yellow, T_m = 69.5 °C) of compound 1 measured by CPM indicates target engagement. D) Compound 1 (magenta sticks, density shown at 5σ) interacts with a pocket formed by the S5, pore helix, and S6. Ser318 of S5 is in position to hydrogen bond (dashed line) with the sulfonamide nitrogen of compound 1. E) Overlay of the K^+ pore of Ca^{2+} -bound SK2-4 (green) and compound 1-bound SK2-4 (yellow) (only 2 subunits are shown for clarity). The methoxy of compound 1 (magenta spheres) clashes with S6 Thr386 (spheres) and induces a movement of the S6 and $S_{45}B$ helices toward the pore axis (arrow) to close the intracellular gate (Val390, spheres). D) Pore radii of Ca^{2+} -bound (green), compound 1-bound (yellow), and compound 4-bound (grey) SK2-4. The location of the intracellular gate (Val390), selectivity filter (SF), and extracellular constriction (Phe243) are indicated. Grey dashed line indicates the radius of hydrated K^+ (3 Å).

benzoxadiazole moiety of compound 1 rests in a hydrophobic pocket lined by Leu321 and Ala325 of S5, Ile352 and Phe356 of the pore helix, and Met381 of S6. In the Ca^{2+} -bound and Ca^{2+} -free apo conformations of SK2-4/CaM this pocket is blocked by Leu321 adopting an alternative rotamer conformation (Fig S6B,C). The central benzamide moiety is positioned between Trp322 of S5 and Phe356 and Leu357 at the C-terminus of the pore helix. The sulfonamide tail of compound 1 extends towards but does not enter the ion pore and the sulfonamide nitrogen hydrogen bonds with Ser318 on S5 (Fig 6D,E). The methoxyphenyl forms an edge-to-face interaction with Phe356 and the methoxy contacts Thr386 on S6, which is one N-terminal helical turn removed from the intracellular gate at Val390 (Fig 6E). The residues that interact with compound 1 are conserved in SK1-3 and thus compound 1 is likely not selective among these isoforms (Fig S1). However, for SK4 Thr212 replaces SK2 Ser318 and Trp216 (homologous to SK2 Trp322) is conserved but adopts a different rotamer conformation (Fig S6D). Both changes occlude the compound 1 binding site in SK4 and would likely reduce compound 1 potency on SK4 as observed in the functional data.

Compared with the ligand-free Ca^{2+} -bound conformation, there are two significant conformational changes in the structure of SK2-4/CaM bound to compound 1. First, there are slight changes in the position of the S5 and pore helix to accommodate the compound 1 core (Fig 6E). These movements translate through Tyr335 and His336 at the extracellular end of S5 to the S3-S4 linker, which is shifted by 1.2 Å in the extracellular direction (Fig 3B, 6E, S6E). In this conformation the hydrophobic constriction at Phe243 would block ion conduction (Fig 6F); however, non-continuous side chain density and a high b-factor (51 Å² compared to 33 Å² in the Ca^{2+} -bound state) indicate increased Phe243 flexibility (Fig S4D). Movement of the S3-S4 linker coincides with conformational changes in the selectivity filter similar to those observed with apamin-bound SK2-4 (described above) resulting in a selectivity filter with four K⁺ coordination sites (Fig S3D, S4D, S6E,F). This structure further supports the hypothesis that movement of the S3-S4 linker away from the selectivity filter induces a conformation with four K⁺ coordination sites.

The second notable conformational change is in the position of the S6 helices. Structural comparisons demonstrate that the methoxy group of compound 1 would clash with Thr386 on S6 in the Ca^{2+} -bound apo conformation of SK2-4 (Fig 6E, S6B). To accommodate the methoxy, Thr386 and the portion of the S6 helices C-terminal to Thr386 shift towards the pore axis akin to the S6 helix shift observed in the Ca^{2+} -free conformation (Fig S6G). Concomitantly, the pore radius at the Val390 intracellular gate is reduced to 1.2 Å demonstrating a clear mechanism for channel inhibition (Fig 6F).

Notably, compound 1 closes the gate at Val390 in the presence of saturating Ca^{2+} concentrations demonstrating that this mechanism of inhibition can override the Ca^{2+} -dependent gating. The importance of the methoxy for inhibition was demonstrated by structurally related compounds 2 and 3, in which the benzoxadiazole of compound 1 is replaced with a benzothiadiazole isostere (Fig S6H,I). Removal of the methoxy from compound 2 produced compound 3 with a 30-fold reduction in potency.

A previous report identified a pair of structurally related compounds that act as selective SK1 inhibitors and activators. This chemical series requires Ser293 on S5 for activity, which corresponds to Leu321 in SK2 suggesting that the SK1 modulators and compound 1 share the same binding site (Fig S1). We hypothesized that compound 1 analogs lacking the methoxyphenyl, which our structure and functional data demonstrated is important for SK2 inhibition, may activate SK2. Indeed, a focused screen of compound 1 analogs lacking the methoxyphenyl identified compound 4 as an SK2 activator (Fig 6A, 7A) which dose-dependently increases SK2 current in the presence of 2 μM intracellular Ca^{2+} . Like compound 1, compound 4 retains the benzoxadiazole/benzamide core, but the benzamide phenyl features a trifluoromethyl substituent para to the amide. In the 3.1 Å structure of Ca^{2+} -bound SK2-4 in complex with compound 4, clear cryo-EM density demonstrates that the benzoxadiazole/benzamide core occupies the same pocket as observed for compound 1 (Fig 7B, S7, Table 1). However, the S6 helices are in an open

conformation and the trifluoromethyl interacts with Ile380, which is 2.5 helical turns N-terminal to Val390, on the S6 helix from a neighboring subunit (**Fig 6F**, 7B). Structural overlays predict that the trifluoromethyl would clash with Ile380 in the closed conformation of the S6 helices and thereby promote the open conformation to activate SK2 (**Fig 7C**). Unlike compound 1, compound 4 binding does not induce movement of the S3-S4 linker and the selectivity filter retains a conformation with two K^+ coordination sites (**Fig 6F**, S7G).

Discussion

The structures of an SK2-4 chimeric channel in the Ca^{2+} -bound and Ca^{2+} -free conformations revealed two unexpected features of the SK2 transmembrane domains. First, the S3-S4 linker forms an anti-parallel β -turn that extends over the S5-S6 segments and interacts with the extracellular pore loops (**Fig 3**). In this conformation, Phe243 on the S3-S4 linker is positioned directly above the selectivity filter creating a hydrophobic constriction with a radius of 1.8 Å, which prevents the efflux of hydrated K^+ ions (**Fig 2D**, 3A,C, 4A). Therefore, the Ca^{2+} -bound SK2-4 structure represents a non-conductive conformation even though the intracellular gate is open. Second, the selectivity filter adopts a conformation that resembles non-selective cation channels, such as NaK and HCN, with only two K^+ coordination sites^{36,39} (**Fig 4A,E**, F, S4A,B). This selectivity filter conformation seems to be due to the interaction between Asp363 at the C-terminus of the selectivity filter and residues Arg240 and Tyr245 on the S3-S4 linker. In the related SK4 channel the homologous aspartate hydrogen bonds with a pore helix tryptophan¹³ (**Fig 4C**), which is critical for the formation of a selectivity filter with four K^+ coordination sites observed in all K^+ selective channels to date^{37,38}. Therefore, by sequestering Asp363 and preventing its interaction with Trp350 in the pore helix, the S3-S4 linker likely induces the two K^+ coordination site selectivity filter observed in the apo SK2-4 structures (**Fig 4A**, S4A,B). In support of this hypothesis, a similar S3-S4 linker and selectivity filter conformation was observed in the structure of rat SK2, which was published during the preparation of this manuscript, and mutagenesis demonstrated that the interactions between the S3-S4 linker and pore residues induce the observed selectivity filter with two K^+ coordination sites³¹. However, this selectivity filter conformation seems to only exist in a non-conductive state of the channel due to the extracellular constriction at Phe243 and thus should not affect the K^+ selectivity observed for SK2 and SK2-4 (**Fig 1B**)^{1,32,36-39,43,44}.

The subsequent structures of SK2-4 bound to apamin and compound 1 elucidated how the SK channels maintain K^+ selectivity in the conductive state. Conductivity likely requires movement of the S3-S4 linker and Phe243 away from the pore axis to expand the extracellular constriction. Such a shift was observed upon binding of both apamin and compound 1 and was accompanied by four structural rearrangements: 1) Asp363 no longer hydrogen bonds with Arg240 in the S3-S4 linker, 2) Asp363 reorients towards Trp350 in the pore helix, 3) Trp350 rotates to enable a hydrogen bond with Asp363, and 4) Tyr361 in the selectivity filter rotates 180° and forms a hydrogen bond with Thr355 in the pore helix (**Fig 5C,D**, S4C,D, S6E,F). These structural changes produce a canonical K^+ -selective selectivity filter with four K^+ coordination sites. Based on these structures we predict that in a conductive state of SK2 the extracellular constriction dilates to allow for the flow of K^+ ions. Such a movement would weaken the interactions between the S3-S4 linker and selectivity filter producing a selectivity filter conformation with four K^+ coordination sites to maintain the K^+ selectivity.

The physiological role of S3-S4 linker and the mechanism of extracellular constriction opening require further investigation. One possibility is that the S3-S4 linker and extracellular constriction exists in an equilibrium between conductive and non-conductive conformations. Such a mechanism may explain some properties of SK2 that are not observed in SK4, which lacks an S3-S4

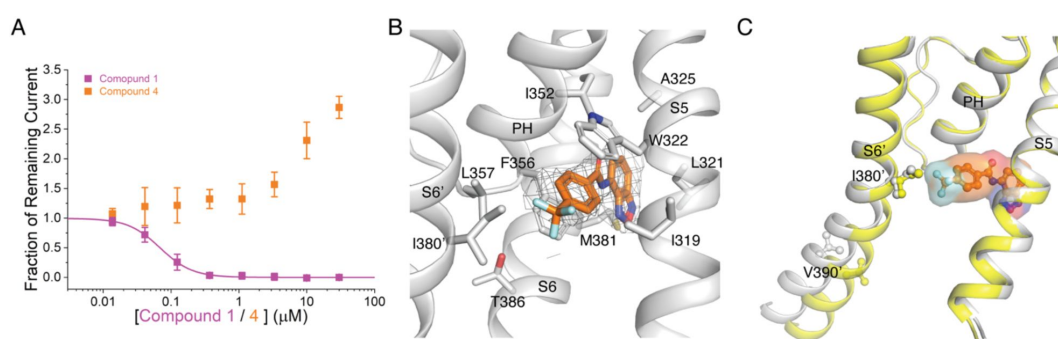


Figure 7

Mechanism of compound 4 activation.

A) Activation or inhibition of SK2 channel by compound 4 (orange squares) and compound 1 (magenta squares), respectively. Change in WT hSK2 current amplitude for different cell populations each exposed to a different extracellular compound concentration was normalized to DMSO control to construct dose responses. Intracellular solution contains 2 μM free Ca^{2+} . SK current was isolated by subtracting leak current determined at the end of experiment by adding saturating concentration of UCL1684, a pore blocker whose potency is not affected by compound 1 or 4. Data points represent mean $\pm$ SD ($n=6$). B) Compound 4 (orange sticks, density shown at 5 σ) interacts with a pocket formed by the S5, pore helix, and S6 and the trifluoromethyl extends towards Ile380 on the neighboring S6 (S6'). C) Overlay of compound 1-bound SK2-4 (yellow) and compound 4-bound (grey) SK2-4. In the closed state of the S6 helices (yellow cartoon) the trifluoromethyl of compound 4 clashes with Ile380 (yellow sticks) on the S6 helix (S6') of the neighboring subunit (2.3 Å distance). In the open state of the S6 helices (grey cartoon), Ile380 (grey sticks) is 3.1 Å distant the trifluoromethyl minimizing this clash.

linker, such as its low conductance (~10 pS) and the ability to switch between low-and high-open probability states.^{1,45} Indeed, mutation of Phe243 in rat SK2 produced a 2-fold increase in channel conductance.³¹

Alternatively, other physiological factors, such as PIP₂^{46,47} or protein-protein interactions,^{48–50} may exist in live cells that modulate the interaction between S3-S4 linker and the selectivity filter. Importantly, the extracellular constriction provides novel opportunities for modulation of SK2.

The structures and functional experiments presented here revealed two binding sites for SK2 modulators and identified three distinct mechanisms of modulation. The bee toxin apamin binds at the extracellular opening of the pore, at the same site as UCL1684, to a site co-formed by the S3-S4 linker from each subunit, explaining why both the S3-S4 linker and pore residues were found to be important for apamin binding and inhibition (**Fig 5**).^{18–20,31} The two essential arginine residues (13 and 14) of apamin are directed towards the selectivity filter and form cation- π interactions with the essential S3-S4 linker residue Phe243, blocking K⁺ ion efflux and inhibiting conduction. In support of this binding interaction and mechanism of inhibition, a Phe243Ala mutant abolishes apamin inhibition (**Fig 5B**). The elucidated apamin binding site and mechanism of inhibition may enable development of new modulators that target the extracellular domains of SK2.

A second partially cryptic binding pocket at the interface of the S5, pore helix, and S6 was revealed through the identification and characterization of compounds 1 and 4 (**Fig 6D**, 7B). This binding pocket interacts with a benzoxadiazole/benzamide core and the functional arm on the benzamide phenyl interacts with the S6 helices to modulate SK2 activity. For compound 1, a sulfonamide methoxyphenyl meta to the amide contacts Thr386 on the S6 helix in the same subunit to induce the closed conformation of the S6 helices and inhibit the channel (**Fig 6D,E**). Conversely, for compound 4 the trifluoromethyl para to the amide contacts Ile380 on the S6 helix from a neighboring subunit to promote the open conformation of the S6 helices and activate the channel (**Fig 6F**, 7B,C). This proposed mechanism of modulation suggests that compound 1 may bind preferentially to the closed conformation of the S6 helices and compound 4 may bind preferentially to the open conformation of the S6 helices. Further optimization of the functional groups on the benzamide phenyl that interact with S6 helices may produce more potent activators and/or inhibitors. For example, more potent activation may be achieved if the interaction between compound 4 and the neighboring S6 was closer to the intracellular gate at Val390 (i.e., within 1 helical turn). Producing isoform selective modulators targeting the compound 1/4 binding pocket may prove more challenging due to the high sequence conservation of residues lining the pocket among SK1-3. However, SK1 modulators that are predicted to share the same binding site as compound 1 take advantage of a unique S5 Ser to confer selectivity over SK2 and SK3 suggesting isoform selectivity could be achieved through further optimization.³⁰

In summary, this study characterized SK2 channel dynamics as well as mechanisms of pharmacological modulation utilizing a SK2-4 chimera, containing the S1-S6 transmembrane regions of human SK2 and the intracellular domains of human SK4.

SK2-4 is K⁺-selective, activated by Ca²⁺, sensitive to SK2 modulators that bind the S1-S6 transmembrane regions, and suitable for cryo-EM structure determination. The findings revealed critical structural features and mechanisms of SK2 channel modulation, providing a framework for development of targeted therapeutics for the SK channel family.

Methods

Construct design

The amino acid sequences for human SK2 (KCNN2, Uniprot Q9H2S1, 579 amino acids) and human SK4 (KCNN4 Uniprot O15554, 427 amino acids) were used to design the chimeric SK2-4 channel construct. The N-terminal portion of SK2 made up of residues 1-123 was replaced with SK4 residues 1-15. The C-terminal portion of SK2 encompassing residues 413-579 was replaced with SK4 residues 306-428. An HRV3C protease recognition sequence followed by a GFP sequence were appended at the C-terminus of SK2-4 resulting in the SK2-4-GFP construct.

The amino acid sequences for SK2-4-GFP and human calmodulin (CaM) (CALM1, Uniprot P0DP23, 149 amino acids) were codon optimized using Twist Bioscience's tool to create DNA sequences for expression. These constructs were synthesized and cloned into the vector pWIL-BacMam for expression and purification. The pWIL-BacMam vector contains the Tn7 transposon and encodes a baculovirus genome with a multiple cloning site, such that when a gene of interest is placed in the pWIL-BacMam vector and transformed into *E. coli* DH10Bac competent cells (ThermoFisher Scientific) it produces bacmid containing the gene of interest. It also contains a CMV enhancer and promoter for expression of the gene of interest in mammalian cells.

SK2-4/CaM chimeric channel complex production

SK2-4 chimeric channel/CaM complex was produced using the BacMam method. Plasmids encoding the constructs (SK2-4-GFP, CaM) were each separately transformed into *E. coli* DH10Bac competent cells (ThermoFisher Scientific) following the manufacturer's protocol & plated onto blue/white selection LB agar plates (Teknova).

After 2 days of incubation at 37 °C, a white colony for each construct was selected and grown overnight in 5 mL liquid medium containing 7 µg/mL gentamicin sulfate, 50 µg/mL kanamycin, and 10 µg/mL tetracycline hydrochloride (Teknova). Cells were harvested by centrifugation at 5000 x g for 5 minutes. Cell pellets were resuspended in 250 µL P1 buffer (Qiagen) and incubated briefly with 250 µL P2 buffer (Qiagen) before addition of 350 µL N3 buffer (Qiagen). Solutions were centrifuged for 10 minutes at 16,000 x g.

Supernatant was added to 1 mL of ice-cold isopropanol and incubated for 20 minutes at -20 °C. Solutions were centrifuged for 15 minutes at 16,000 x g and 4 °C to pellet DNA. Supernatant was discarded and pellet was washed with 1 mL of ice-cold 70% ethanol, followed by centrifugation at 16,000 x g for 10 minutes. Supernatant was discarded and pellet was left to air-dry for 10 minutes before resuspension with 50 µL of molecular-biology grade water.

Baculovirus for each construct was produced by transfection of bacmid into Sf9 cells as follows: 3 µL X-tremeGENE HP DNA transfection reagent (Roche), 5 µL of bacmid, and 100 µL transfection medium (Expression Systems) were incubated for 15 minutes at room temperature. After incubation, 2.5 mL of Sf9 insect cells (Expression Systems) at a density of 1 million cells/mL were added. Cells were incubated for 1 week with shaking at 27 °C before baculovirus-containing medium was harvested. Baculovirus was amplified in Sf9 cells for two additional rounds after transfection. Expi293F cells (ThermoFisher Scientific) were grown in suspension at 37 °C in 8% CO₂ atmosphere with 110 rpm shaking to a density of approximately 3 x 10⁶ cells/mL before transduction with virus. Virus was added to the culture with 2% v/v CaM baculovirus and 8% v/v of SK2-4-GFP baculovirus. Approximately 16 hours after transduction, sodium butyrate was added to cultures to a final concentration of 10 mM. Cultures were incubated an additional ~48 hours at 37 °C in 8% CO₂ atmosphere with 110 rpm shaking before harvest by centrifugation at 6,000 x g for 45 minutes. Cell pellets were flash-frozen in liquid nitrogen and stored at -80 °C until use.

Cell pellet from a 4 L culture was resuspended in 200 mL of room-temperature resuspension buffer (10 mM Tris pH 8, 20 mM KCl, 2 mM CaCl_2 , 0.5 mM MgCl_2 , 0.05 mg/mL DNase I, Pierce protease inhibitor tablet EDTA-free) for 30 minutes with vigorous stirring. Lysate was centrifuged for 45 minutes at 35,500 x g to collect membranes. The membrane pellet was resuspended in 200 mL of extraction buffer (10 mM Tris pH 8, 20 mM KCl, 2 mM CaCl_2 , 2% w/v n-dodecyl- β -D-maltopyranoside (DDM), 0.4% w/v cholesteryl hemisuccinate tris salt (CHS), Pierce protease inhibitor tablet EDTA-free) using a dounce. Membranes were solubilized with vigorous stirring at 4 °C for 2 hours. Solution was clarified by ultracentrifugation at 195,000 x g for 1 hour at 4 °C.

Supernatant was incubated with 4 mL GFP nanobody resin (Bulldog Bio) pre-equilibrated with wash buffer (20 mM Tris pH 8, 150 mM KCl, 2 mM CaCl_2 , 0.05% w/v DDM, 0.01% w/v CHS) for 1.5 hours at 4 °C with gentle agitation. Resin was collected over a gravity column and washed with 5 column volumes (CV) of wash buffer supplemented with 5 mM ATP and 10 mM MgCl_2 , followed by washing with 10 CV of unsupplemented wash buffer. Resin was resuspended in 20 mL of wash buffer with 0.25 mg of HRV3C protease and incubated overnight at 4 °C with gentle agitation to cleave untagged SK2-4/CaM complex from the resin. Supernatant containing protein was collected and concentrated to 0.5 mL using 100 kDa MWCO Amicon Ultra centrifugal filter (EMD Millipore).

Concentrated protein was applied to a Superose 6 Increase 10/300 GL column (Cytiva) pre-equilibrated with SEC buffer. For samples purified in presence of Ca^{2+} , the SEC buffer was 20 mM Tris pH 8, 150 mM KCl, 2 mM CaCl_2 , 0.005% w/v glyco-diosgenin (GDN), 0.0005% w/v CHS. For Samples purified in absence of Ca^{2+} , the SEC buffer was 20 mM Tris pH 8, 150 mM KCl, 5 mM EGTA, 0.005% w/v GDN, 0.0005% w/v CHS. Size exclusion was performed at 4 °C with a flow rate of 0.5 mL/minute. SDS-PAGE was performed using NuPAGE Bis-tris 4-12% gel (Invitrogen) and NuPAGE MES SDS running buffer (Invitrogen) to identify fractions with pure SK-4/CaM complex. Fractions were collected and concentrated using 100 kDa MWCO Amicon Ultra centrifugal filter to desired concentration and frozen directly onto cryo-EM grids.

EM sample preparation and data collection

SK2-4/CaM complex samples were prepared at the following concentrations: Ca^{2+} -bound SK2-4/CaM at 7.5 mg/mL; Ca^{2+} -free SK2-4/CaM at 8.25 mg/mL. For samples with apamin present, Ca^{2+} -bound SK2-4/CaM was supplemented with apamin (Sigma-Aldrich) to a final concentration of 200 μM . For the sample with compound 1, Ca^{2+} -bound SK2-4/CaM was supplemented with compound 1 in 100% DMSO to a final concentration of 200 μM compound 1 and 2% DMSO. For the sample with compound 4, Ca^{2+} -bound SK2-4/CaM was supplemented with compound 4 in 100% DMSO to a final concentration of 500 μM compound 4 and 2% DMSO.

UltrAuFoil® R 0.4/1.2 200 mesh grids were used for the Ca^{2+} -bound and Ca^{2+} -free SK2-4/CaM samples and UltrAuFoil® R 1.2/1.3 300 mesh grids were used for all other samples. The grids (Electron Microscopy Sciences) were glow-discharged for 30 seconds at 0.5 mBar with 15 mA using easiGlow system (PELCO). 5 μL of sample were applied to grids at 4°C with 100% humidity. After 30 seconds, grids were blotted for 5 seconds with blot force 25 and plunged into liquid ethane using the Vitrobot Mark IV system (Thermo Fisher).

For Ca^{2+} -bound SK2-4/CaM, Ca^{2+} -free SK2-4/CaM, SK2-4/CaM + Apamin, and Ca^{2+} -bound SK2-4/CaM + compound 1, data were collected on a Titan Krios microscope (Thermo Fisher Scientific, Waltham, MA, USA) equipped with a Cs-corrector hardware operated at an accelerating voltage of 300 kV with a 50 μm C2 aperture at an indicated magnification of 75,000 $\times$ in nanoprobe mode. A Falcon4i Direct Electron Detector camera operated in Electron-Event representation (EER) mode was used to acquire dose-fractionated images with Thermo Fisher Scientific EPU software. For Ca^{2+} -bound SK2-4/CaM + compound 4, data were collected on a Glacios microscope (Thermo Fisher Scientific, Waltham, MA, USA) operated at an accelerating voltage of 200 kV with a 50 μm C2

aperture at an indicated magnification of 120,000 $\times$ in nanoprobe mode. A Falcon3 Direct Electron Detector camera was used to acquire dose-fractionated images with Thermo Fisher Scientific EPU software. Data collection parameters for each data set are listed in table S1.

EM data processing

[Ca²⁺-bound SK2-4/CaM] Movie stacks were gain-corrected and motion-corrected in patches using MotionCor2⁵¹, Motion-corrected images were imported into cryoSPARC4⁵². Contrast transfer function parameters were determined using Patch CTF. Blob picking was performed with a subset of the micrographs with an expected diameter of 160 to 180 Å. Selected particles were extracted with a box size of 360 pixels and subjected to reference-free 2D classification to generate templates, which were used for subsequent template-based autopicking with the full set of micrographs. 8,310,008 particles were selected and extracted with a box size of 360 pixels. After iterative rounds of reference-free 2D classification 910,204 particles were used to generate two *ab initio* models. The 658,368 particles from the better model were subjected to non-uniform 3D refinement⁵³ with imposed C4 symmetry using the *ab initio* model as a template resulting in a 3.1 Å consensus refinement. After local CTF refinement, a soft mask excluding the detergent micelle was generated for local refinement with imposed C4 symmetry, resulting in the final consensus refinement with a resolution of 3.1 Å.

[Ca²⁺-free SK2-4/CaM] Movie stacks were gain-corrected and motion-corrected in patches using MotionCor2⁵¹, Motion-corrected images were imported into cryoSPARC4⁵². Contrast transfer function parameters were determined using Patch CTF. Blob picking was performed with a subset of the micrographs with an expected diameter of 160 to 180 Å. Selected particles were extracted with a box size of 360 pixels and subjected to reference-free 2D classification to generate templates, which were used for subsequent template-based autopicking with the full set of micrographs.

8,160,096 particles were selected and extracted with a box size of 360 pixels. After iterative rounds of reference-free 2D classification 386,845 particles were subjected to non-uniform 3D refinement⁵³ with imposed C4 symmetry using the Ca²⁺-bound SK2-4/CaM consensus refinement as a template, resulting in a 3.4 Å consensus refinement. A soft mask excluding the detergent micelle and the intracellular stalk was generated for local refinement with imposed C4 symmetry, resulting in the final consensus refinement with a resolution of 3.4 Å.

[SK2-4/CaM + Apamin] Movie stacks were gain-corrected and motion-corrected in patches using MotionCor2⁵¹, Motion-corrected images were imported into cryoSPARC4⁵². Contrast transfer function parameters were determined using Patch CTF. Template autopicking was performed using the 2D class averages from the SK2-4/CaM + Ca²⁺ data set. 7,528,863 picked particles were extracted using a box size of 360 pixels. Iterative rounds of reference-free 2D classification led to 1,440,270 particles which were used to generate two *ab initio* models. The better model comprising of 1,186,640 particles was subjected to non-uniform refinement⁵³ using the *ab initio* model as a template, first with C4 symmetry imposed, which led to a consensus refinement at 3 Å. Density at the top of the SK2-4 pore was identified as the likely apamin binding site. A soft ovoid mask covering the top of the pore and the SK2-4 selectivity filter was used for focused 3D classification without alignment of the particles with C1 symmetry and five classes. Three classes comprising of 734,192 total particles had clear asymmetrical density for apamin at the extracellular opening of the pore. The classes appeared to be structurally identical but rotated by 90° from one another. Volume alignment tools were used to rotate the volumes and associated particle stacks 90° or 180° around the symmetry axis so the three classes were in the same orientation. These particle stacks were put into a local refinement with C1 symmetry using one of the classes as a template and with a soft mask that excluded the detergent micelle and intracellular stalk of SK2-4, resulting in a consensus refinement at 3.1 Å with clear density for apamin at the extracellular pore opening. This was followed by a non-uniform refinement with C1 symmetry with the local refinement map used as a template, resulting in a final consensus refinement at 3.2 Å.

[Ca²⁺-bound SK2-4/CaM + compound 1] Movie stacks were gain-corrected and motion-corrected in patches using MotionCor2⁵¹. Motion-corrected images were imported into cisTEM⁵⁴ for contrast transfer function estimation followed by particle picking. 1,884,323 selected particles were extracted with a box size of 360 pixels and subsequently imported into cryoSPARC4⁵². Reference-free 2D classification was performed to select 326,099 good particles. These were used to generate two *ab initio* models. The better model comprising of 262,274 particles was subjected to non-uniform refinement⁵³ using the Ca²⁺-bound SK2-4/CaM model as a template, resulting in a consensus refinement at 3.3 Å. This refinement was put through global CTF refinement, followed by another round of non-uniform refinement resulting in the final map at 3.3 Å.

[Ca²⁺-bound SK2-4/CaM + compound 4] Movie stacks were gain-corrected and motion-corrected in patches using MotionCor2⁵¹. Motion-corrected images were imported into cryoSPARC4⁵². Contrast transfer function parameters were determined using Patch CTF. Template-based autopicking, extraction with a box size of 360 pixels, and reference-free 2D classification were performed to select 436,179 good particles. These were used to generate two *ab initio* models. The better model comprising of 360,727 particles was subjected to non-uniform refinement⁵³ using the Ca²⁺-bound SK2-4/CaM model as a template, resulting in a consensus refinement at 3.1 Å.

Model building

For the Ca²⁺-bound SK2-4/CaM structure, the SK2-4 region was built *de novo* in Coot⁵⁵ using the 3.1 Å-EM density map. For CaM, a chain of CaM from the SK4 structure 6CNN was docked into the cryo-EM map in Chimera⁵⁶, followed by adjustment in Coot. 4-fold symmetry was enforced on the model in Coot. The model was refined against the cryo-EM density map using iterative rounds of manual adjustment in Coot followed by real space refinement in Phenix⁵⁷.

For the model building of the Ca²⁺-free SK2-4/CaM, apamin-bound, compound 1-bound, and compound-4 bound structures the Ca²⁺-bound SK2-4/CaM structure was used as a starting point and manually docked into the corresponding cryo-EM density maps, followed by manual adjustment in Coot⁵⁵. For the apamin-bound structure, apamin from the NMR structure 7OXF was docked into the cryo-EM density map using Chimera⁵⁶, followed by truncation of unresolved regions and manual adjustment in Coot. For the Ca²⁺-free, compound 1-bound, and compound 4-bound structures, 4-fold symmetry was enforced in Coot. No symmetry was enforced for the apamin-bound model. The models were refined against their respective cryo-EM density maps using iterative rounds of manual adjustment in Coot followed by real space refinement in Phenix⁵⁷.

For all models, if the backbone was not visible in the density map, the residue was not modeled. Residues with poorly-defined sidechain density were truncated at Cβ. The quality of the models was determined using MolProbity⁵⁸. Figures were prepared using PyMOL (Schrödinger, LLC, 2010. The PyMOL Molecular Graphics System, Version 3.1.3) and Chimera⁵⁶.

Electrophysiology assays

SK channel assays

All electrophysiology assays for SK used CHO-K1 cells either stably expressing SK channels (WT SK2 and SK4) or transiently transduced with channel baculovirus constructs (WT & mutant SK2, SK4 and SK2-4 chimera) using the BacMam system. Stably channel-expressing cells were cultured using T150 flasks in 37 °C incubator with 5% CO₂ until 70-80% confluency. Culture media were composed of Dulbecco's modified Eagle's medium/F-12 nutrient mixture (DMEM/F-12, #31320, Gibco) and 10% fetal bovine serum (#89510-196, Avantor), 1% penicillin/streptomycin (#15140, Gibco) and 1% non-essential amino acids (#11140, Gibco). For BacMam transduction, 8 million parental CHO-K1 cells were seeded into a T150 flask with 15 mL media. After >6 hours in 37 °C

incubator with 5% CO₂ and until cells were attached, media were aspirated and replaced with 10 mL fresh media containing 400 µL baculovirus for individual channel constructs. After this, cells were cultured in 30 °C incubator with 5% CO₂. After overnight media were aspirated and replaced with 10 mL media containing 4µM Tricostatin A (#T-1952, Sigma) for 4 hours before dilution to 1 µM by adding 30 mL media. Cells were ready to use >40 hours in 30 °C incubator after transduction. On the day of experiments for all cells, media were aspirated before a wash with PBS (#14190, Gibco). Cells were then dissociated using 5 mL Tryple (#12605010, ThermoFisher) for 3 min at 37 °C, spun down and resuspended in serum free medium (#12052114, Gibco) with added 25 mM HEPES (#15630, Gibco). Right before experiments, cells were spun down again and resuspended in extracellular buffer containing 5 mM BaCl₂ at 2-3 million cells/mL before loaded to the “cell transfer plate” of the Qube instrument (Sophion).

All SK electrophysiological recordings were made with the automated Qube system and 384X 10-hole QChips at 22 °C. Sum of SK currents from up to 10 cells are measured in each recording well from the 384-well plate. Briefly, cells from Qube “cell transfer plate” were pipetted into each well to form seals then broken into whole-cell configuration.

Extracellular buffer contains (in mM): 130 NaCl, 6 KCl, 2 CaCl₂, 40 sucrose, 10 HEPES, 1 MgCl₂, pH=7.4 with NaOH. To enhance sealing, 5 or 15 mM BaCl₂ were added to extracellular buffer (sucrose reduced accordingly to maintain osmolarity) which were washed away with Ba²⁺ free extracellular buffer after seal was formed. Intracellular buffer contains (in mM): 110 K₂SO₄, 8 NaCl, 4.68 CaCl₂, 10 HEPES, 5 EGTA, 5 HEDTA, and 4 Mg-ATP added right before experiments, pH=7.0 with KOH. All SK experiments use this intracellular solution which contains 2 µM free Ca²⁺ (determined by online Maxchelator program at <https://somapp.ucdmc.ucdavis.edu/pharmacology/bers/maxchelator/webmaxc/webmaxc> E.htm), except for one case when intracellular free Ca²⁺ was varied to evaluate Ca²⁺ dependent activation as mentioned below. SK current was recorded using a 250 ms long voltage ramp from -120 mV to +80 mV, whereas current level at ~0 mV was taken as the current amplitude to minimize contribution by leak. No series resistance compensation or leak subtraction was included in voltage protocol. For pharmacological effects, cells were washed with Ba²⁺ free extracellular buffer containing different concentrations of compounds before SK currents were recorded (a single concentration per well). Leak and non-SK currents were estimated at the end of experiment with saturating concentration of specific SK inhibitors (10 µM UCL1684, 30 µM AP14145, or 10 µM TRAM-34, as needed) and subtracted from average current amplitudes. Average current amplitude after compound from each well was normalized to its before-compound amplitude then normalized to time-matched wells treated with 0.3% DMSO. Normalized values at each concentration (n=4-6 wells) were used to construct dose response curves, which were fitted with Hill Equation to yield IC₅₀ values. For Ca²⁺ dependent activation of SK channels, different amounts of CaCl₂ were added to intracellular buffer to achieve desired levels of free Ca²⁺ concentration as calculated with Maxchelator. On a 384-well plate, 24 wells (3 columns) were exposed to intracellular buffer containing one of eight different calculated free Ca²⁺ concentrations ranging from 0.1 to 20 µM. Average SK current amplitudes from 24 wells for each [Ca²⁺] were normalized to the level at 20 µM and plotted against free Ca²⁺ concentration to construct the Ca²⁺ dependent activation curve, which was fitted with the Hill Equation to yield EC₅₀ values. Note that on the automated high throughput electrophysiology system, it is only feasible to record current in whole-cell configuration (in contrast to inside-out) in which true intracellular free [Ca²⁺] may differ from the calculated values for the bulk intracellular buffers due to incomplete diffusion.

Counter-screening assays on other ion channels

Compound effect on hERG channel was tested on the Qube system at 35 °C using CHO-K1 cells stably expressing hERG channel. Assay conditions were similar as the SK assay except for buffers and voltage protocol. Extracellular buffer contains (in mM): 145 NaCl, 4 KCl, 2 CaCl₂, 10 glucose, 10 HEPES, 1 MgCl₂, pH=7.4 with NaOH. Intracellular buffer contains (in mM): 20 KCl, 120 KF, 10 EGTA,

10 HEPES, pH=7.2 with KOH. hERG channel inhibition was measured by peak tail current level at -50 mV after a 3.5-sec long depolarization to 10 mV. Compound effect on hCav1.2 channel was tested on the Qube system at 35 °C using CHO-K1 cells expressing hCav1.2 channel (inducible with the T-Rex system). Assay conditions were similar as the SK assay except for buffers and voltage protocol. Extracellular buffer contains (in mM): 145 NaCl, 10 CaCl₂, 10 HEPES, 4 KCl, pH=7.4 with NaOH. Intracellular buffer contains (in mM): 112 CsCl, 28 CsF, 2 NaCl, 10 HEPES, 8.2 EGTA, pH=7.3 with CsOH. Cav1.2 channel inhibition was measured by peak current level at 10 mV depolarization from a -70 mV holding potential. Compound effect on hKCNQ1 channel was tested on the Qube system at 22 °C using CHO-K1 cells stably expressing hKCNQ1 channel. Assay conditions were similar as the SK assay except for voltage protocol. Extracellular buffer contains (in mM): 130 NaCl, 6 KCl, 2 CaCl₂, 40 sucrose, 10 HEPES, 1 MgCl₂, pH=7.4 with NaOH. Intracellular buffer contains (in mM) 110 K₂SO₄, 8 NaCl, 4.68 CaCl₂, 10 HEPES, 5 EGTA, 5 HEDTA and 4 Mg-ATP added right before experiments, pH=7.0 with KOH. hKCNQ1 channel inhibition was measured by peak current level at 4-sec long depolarization to 50 mV from a -80 mV holding potential.

Compound effect on hNav1.5 channel was tested on the QPatch (Sophion) system at room temperature using CHO-K1 cells stably expressing hNav1.5 channel. QPatch is a different automated electrophysiology system than the Qube and uses 48-well plates. Extracellular buffer contains (in mM): 137 NaCl, 4 KCl, 1 MgCl₂, 1.8 CaCl₂, 10 HEPES, 10 Glucose, pH=7.3 with NaOH. Intracellular buffer contains (in mM): 120 CsF, 20 CsCl, 10 HEPES, 5 EGTA, 10 NaCl, pH=7.2 with CsOH. hNav1.5 channel inhibition was measured by steady-state peak current level at -10 mV depolarization of 400 ms at 1 Hz from a holding potential of -80 mV. In contrast to Qube assays, compound effect was tested by perfusing individual cells with compounds of increasing concentrations to construct cumulative dose response curves. All dose response data points were normalized to time matched wells / cells treated with DMSO. Dose response curves were plotted as mean $\pm$ SD and fitted with the Hill Equation.

Differential Scanning Fluorimetry assay

Differential scanning fluorimetry (DSF) assay was performed in DSF assay buffer (20 mM Tris pH 8, 150 mM KCl, 2 mM CaCl₂, 0.005% w/v GDN, 0.0005% w/v CHS). SK2-4/CaM complex in DSF assay buffer was prepared at a concentration of 0.042 mg/mL. compound 1 was dissolved in 100% DMSO at 10 mM before being diluted into the DSF assay buffer to 200 μ M, with a final DMSO concentration of 2%. A negative control solution was prepared in the same way using only DMSO in the absence of compound. 7-Diethylamino-3-(4'-Maleimidylphenyl)-4-Methylcoumarin (CPM) was dissolved in 100% DMSO to a concentration of 4 mg/mL before being diluted into the DSF assay buffer to a concentration of 0.2 mg/mL.

Assay wells were prepared in triplicate by adding 30 μ L of the SK2-4/CaM solution, 30 μ L of the compound 1 solution or negative control solution, and 5 μ L of the CPM solution together in Mx300P 96-well non-skirted plates (Agilent Technologies) such that each well contained 1.25 μ g of SK2-4/CaM complex and 100 μ M of compound 1, if present.

Combined solution was briefly mixed followed by centrifugation. Plate was analyzed using Stratagene Mx3005P (Agilent Technologies) with the following method: temperature was held at 25 °C for 30 minutes, followed by an increase of 0.5 °C every 30 seconds until the temperature reached 95 °C. Fluorescence readings were taken at each temperature.

Raw data were analyzed by fitting melt curves using an extended Boltzmann function to determine melting temperature T_m for each replicate. Reported ΔT_m was determined by taking the mean of the three replicates for the compound 1 samples and subtracting the mean of the three replicates for the apo samples.

Chemical synthesis of compound 1 (N-(benzo[c][1,2,5]oxadiazol-4-yl)-3-((4-methoxyphenyl)sulfonamido)benzamide) and compound 4 (N-(benzo[c][1,2,5]oxadiazol-4-yl)-4-(trifluoromethyl)benzamide)

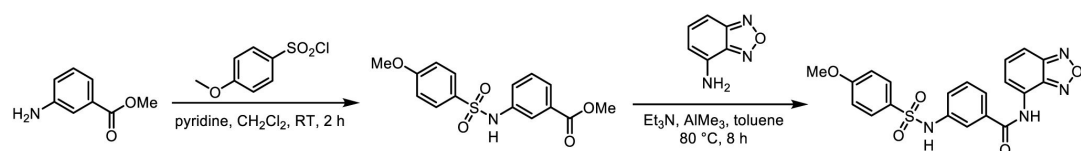
General methods

Unless otherwise noted, reagents and solvents were used as received from commercial suppliers. Compounds 2 and 3 were purchased from commercial suppliers. Proton nuclear magnetic resonance (NMR) spectra were obtained on either a Bruker Avance spectrometer or a Varian Oxford 400 MHz spectrometer unless otherwise noted. NMR spectra are given in ppm (δ) and coupling constants, J , are reported in Hertz.

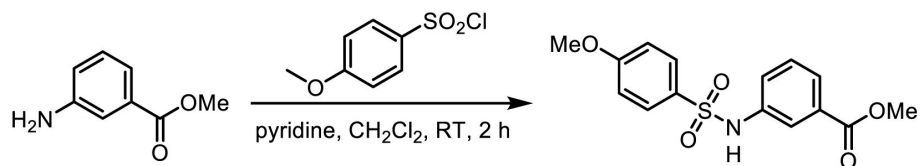
Tetramethylsilane (TMS) was used as an internal standard. Chemical shifts are reported in ppm relative to dimethyl sulfoxide (δ 2.50), methanol (δ 3.31), chloroform (δ 7.26) or other solvent as indicated in NMR spectral data. Peaks are reported as (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet or unresolved, br = broad signal, coupling constant(s) in Hz, integration). ^{13}C NMR spectra were recorded with ^1H -decoupling on Bruker AV 101 MHz spectrometers and are reported in ppm with the solvent resonance employed as the internal standard (DMSO- d_6 at 39.52 ppm). Mass spectra (ESI-MS) were collected using a Waters System (Acquity UPLC and a Micromass ZQ mass spectrometer) or Agilent-1260 Infinity (6120 Quadrupole); all masses reported are the m/z of the protonated parent ions unless recorded otherwise. The chemical names were generated using ChemDraw Professional v23 from Perkin Elmer Informatics.

Temperatures are given in degrees Celsius. If not mentioned otherwise, all evaporations are performed under reduced pressure, typically between about 15 mm Hg and 100 mm Hg (= 20-133 mbar).

Synthetic scheme for N-(benzo[c][1,2,5]oxadiazol-4-yl)-3-((4-methoxyphenyl)sulfonamido) benzamide (1)



Synthesis of methyl 3-((4-methoxyphenyl)sulfonamido)benzoate



To a stirred solution of methyl 3-aminobenzoate (2.0 g, 1.0 equiv, 13.2 mmol) in DCM (10.0 mL) was added pyridine (3.14 g, 3.21 mL, 3 equiv, 39.7 mmol), followed by dropwise addition of 4-methoxybenzenesulfonyl chloride (3.28 g, 1.2 equiv, 15.6 mmol) at 25 °C. Upon complete addition, the reaction mixture was stirred at RT for 2 h. The progress of the reaction was monitored by TLC and LCMS. Upon completion of the reaction, the reaction mixture was diluted with water and extracted with DCM twice. The combined organic layers were washed with ice water twice,

washed with brine, dried over Na_2SO_4 , filtered, and concentrated under reduced pressure to obtain a residue. The crude residue was triturated with n-pentane and diethyl ether to get methyl 3-((4-methoxyphenyl)sulfonamido)benzoate (3.0 g, 9.1 mmol, 69% yield) as a brown solid.

LCMS

System: Shimadzu-LCMS 2020 (single quad)

Column: ACQUITY UPLC BEH C18 1.7 μm , 2.1*50 mm

Column temperature: 40 °C

Gradient (Time (min.) / %B): 0.01/2, 0.08/2, 1.50/50, 2.20/98, 3.60/98, 4.20/2, 5.00/2

Eluent A: 0.1% HCOOH in water

Eluent B: 0.1% HCOOH in CH_3CN

Flow: 0.8 mL/min

ESI-MS, negative mode, m/z 320.1 $[\text{M}-\text{H}]^-$, Rt: 3.25 min, 99%.

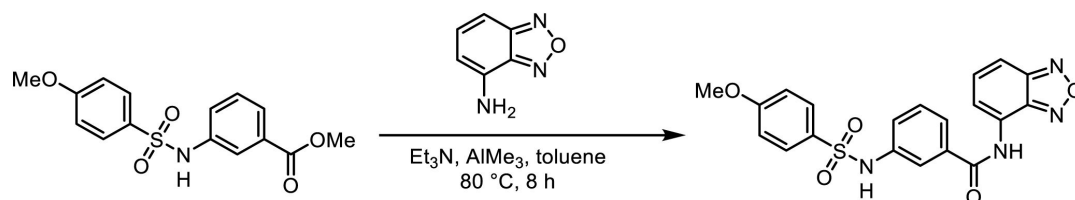
HPLC

Column: PRUDENT C18 150X4.6 mm, 5 μm Flow: 1.0 mL/min

Mobile phase: (A) 0.01% TFA in water, (B) ACN Gradient: T/%B 0/30, 1/70, 6/100, 8/100, 10/30, 12/30
Rt: 4.88 min, 98%

^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ = 10.42 (s, 1H), 7.71 - 7.68 (m, 3H), 7.63 - 7.57 (m, 1H), 7.41 - 7.35 (m, 2H), 7.10 - 7.0 (m, 2H), 3.79 (s, 3H), 3.76 (s, 3H).

Synthesis of N-(benzo[c][1,2,5]oxadiazol-4-yl)-3-((4-methoxyphenyl)sulfonamido) benzamide (1)



To a stirred mixture of methyl 3-((4-methoxyphenyl)sulfonamido)benzoate (500 mg, 1.0 equiv, 1.56 mmol) in toluene (5 mL) was added triethylamine (315 mg, 434 μL , 2 equiv, 3.11 mmol) and benzo[c][1,2,5]oxadiazol-4-amine (252 mg, 1.2 equiv, 1.87 mmol) followed by dropwise addition of trimethylaluminum (2M in hexane) (336 mg, 447 μL , 3 equiv, 4.67 mmol) at RT. After complete addition the reaction mixture was heated at 80

°C for 8 h. The progress of the reaction was monitored by LCMS and TLC. The reaction mixture was cooled to RT and quenched with ice water and extracted with ethyl acetate twice. The combined organic layers were washed with brine, dried over Na_2SO_4 , filtered and concentrated to obtain a crude residue. The crude product was purified by normal phase silica gel chromatography, eluting with 10-25% ethyl acetate in hexane. The pure fractions were pooled and concentrated to obtain

the product as a dark yellow gummy mass which was triturated with n-pentane and diethyl ether to afford N- (benzo[c][1,2,5]oxadiazol-4-yl)-3-((4-methoxyphenyl)sulfonamido)benzamide (**1**) (260 mg, 0.60 mmol, 39% yield) as a yellow solid.

LCMS

System: Shimadzu–LCMS 2020 (single quad)

Column: ACQUITY UPLC BEH C18 1.7 μ m, 2.1*50 mm

Column temperature: 40 °C

Gradient (Time (min.) / %B): 0.01/2, 0.08/2, 1.50/50, 2.20/98, 3.60/98, 4.20/2, 5.00/2

Eluent A: 0.1% HCOOH in water Eluent B: 0.1% HCOOH in CH₃CN

Flow: 0.8 mL/min

ESI-MS *m/z* 424.9, [M+H]⁺, Rt: 2.92 min, 99%

HPLC

Column: X BRIDGE C18 150X4.6 mm, 3.5 μ m

Flow: 1.0 mL/min

Mobile phase: (A) 0.1% FORMIC ACID IN WATER, (B) ACN

Gradient: T/%B 0/5,1/5, 6/100, 8/100, 10/5, 12/5 Column Temperature: 40 °C

Rt: 7.03 min, 98.36%

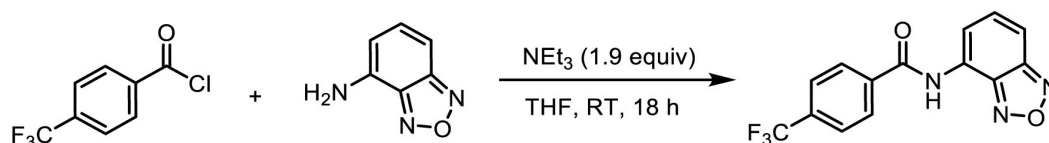
¹H NMR (400 MHz, DMSO-*d*₆) δ 10.82 (s, 1H), 10.43 (s, 1H), 7.87 – 7.81 (m, 2H), 7.76

– 7.67 (m, 4H), 7.64 (dd, *J* = 9.1, 7.1 Hz, 1H), 7.46 – 7.39 (m, 1H), 7.37 – 7.32 (m, 1H), 7.10 – 7.03 (m, 2H), 3.79 (s, 3H).

¹³C NMR (101 MHz, DMSO-*d*₆) δ 165.6, 162.5, 149.6, 145.6, 138.4, 134.7, 133.5, 130.9, 129.3, 128.9, 126.7, 123.1, 123.0, 122.0, 119.5, 114.5, 111.8, 55.6.

HRMS (ESI) *m/z* calculated for C₂₀H₁₇N₄O₅S⁺ [M+H]⁺ 425.0914, found 425.0904.

Synthetic scheme for N-(benzo[c][1,2,5]oxadiazol-4-yl)-4- (trifluoromethyl)benzamide (**4**)



Synthesis of N-(benzo[c][1,2,5]oxadiazol-4-yl)-4-(trifluoromethyl)benzamide (4)

To a solution of benzo[c][1,2,5]oxadiazol-4-amine (51 mg, 0.38 mmol, 1.0 equiv) in THF (1.75 mL) in a 10 mL glass vial was added 4-(trifluoromethyl)benzoyl chloride (100 mg, 0.48 mmol, 1.25 equiv) followed by triethylamine (73 mg, 0.72 mmol, 1.9 equiv). The resulting mixture was shaken in a shaker for 18 hours. Then, the mixture was filtered and concentrated *in vacuo*. The residue was re-dissolved in methanol/acetonitrile (3 mL) and directly subjected to purification by HPLC. The product-containing fractions were pooled and concentrated *in vacuo* to furnish N-(benzo[c][1,2,5]oxadiazol-4-yl)-4-(trifluoromethyl)benzamide (4) (50 mg, 0.16 mmol, 43% yield) as a light yellow solid.

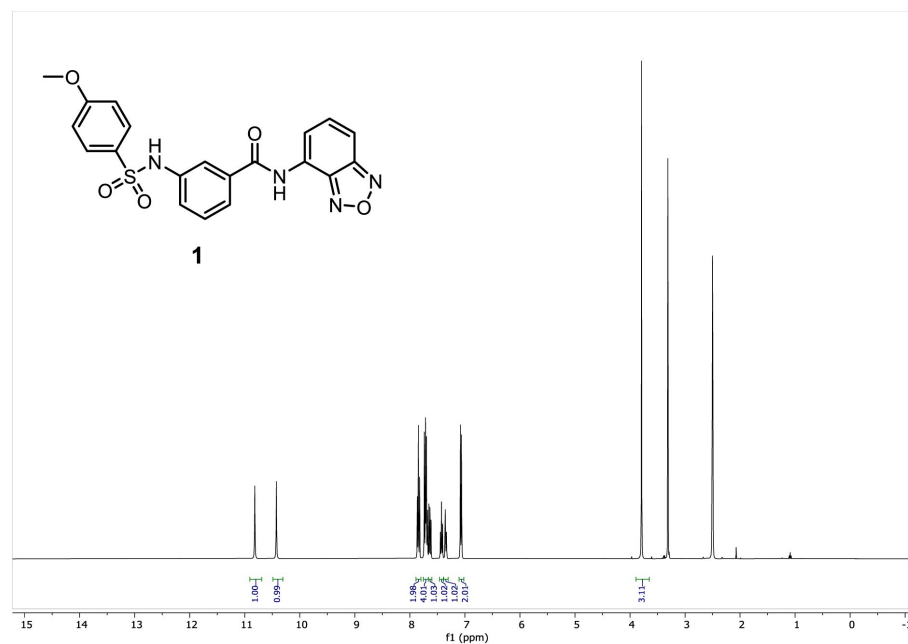
¹H NMR (400 MHz, DMSO-*d*₆) δ 11.15 (s, 1H), 8.23 – 8.18 (m, 2H), 7.98 – 7.93 (m, 3H), 7.88 – 7.85 (m, 1H), 7.67 (dd, *J* = 9.1, 7.1 Hz, 1H).

¹³C NMR (101 MHz, DMSO-*d*₆) δ 165.2, 149.6, 145.6, 137.5, 133.5, 131.7 (q, *J* = 32.2 Hz), 129.0, 126.4, 125.5 (q, *J* = 3.7 Hz), 123.9 (d, *J* = 272.8 Hz), 122.2, 112.1.

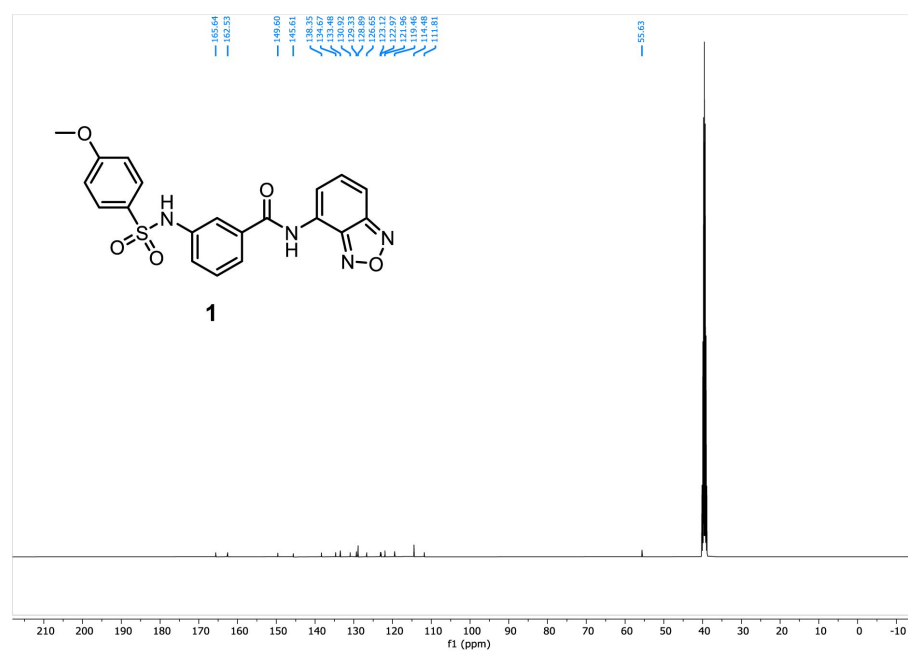
¹⁹F NMR (377 MHz, DMSO-*d*₆) δ –61.4.

HRMS (ESI) *m/z* calculated for C₁₄H₉F₃N₃O₂⁺ [M+H]⁺ 308.0641, found 308.0650.

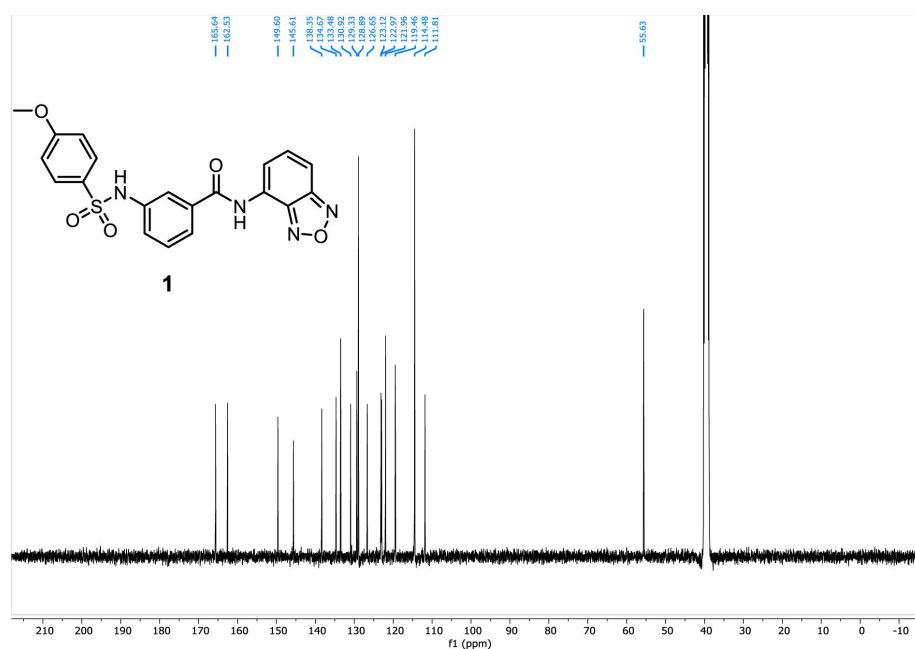
¹H NMR spectrum of compound 1 in DMSO-*d*₆



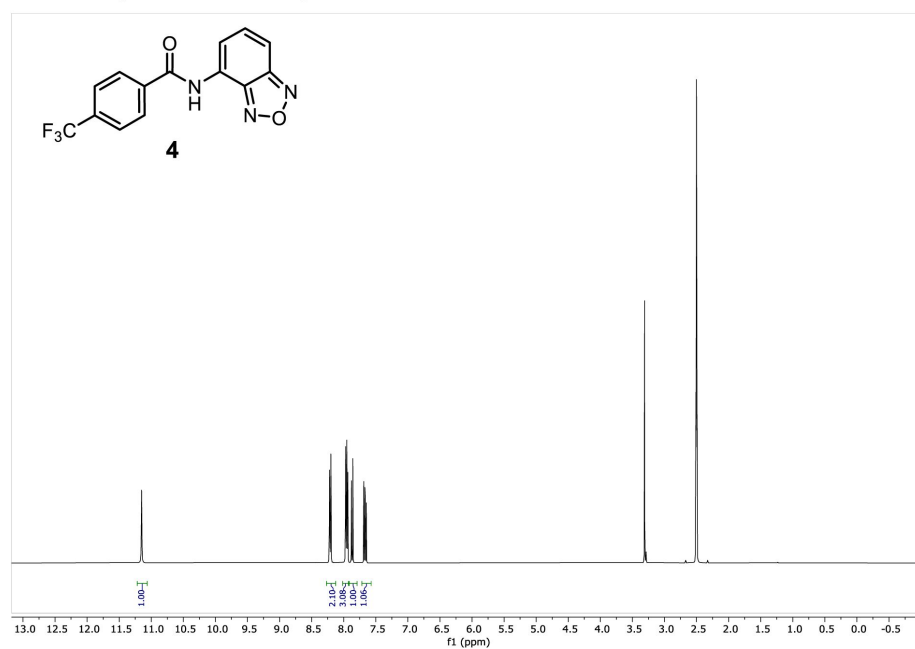
¹³C NMR spectrum of compound 1 in DMSO-*d*₆



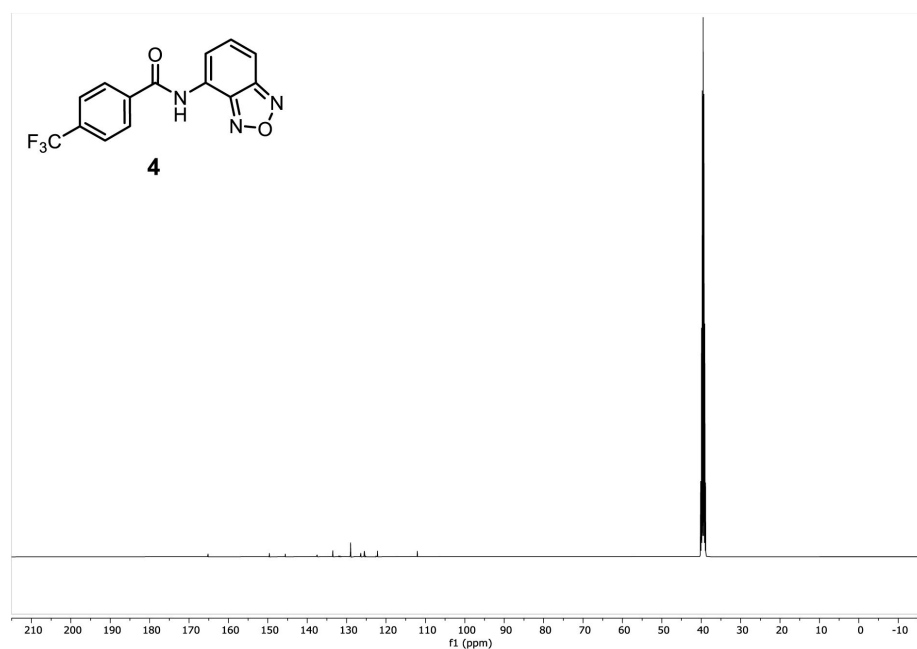
¹³C NMR spectrum of compound 1 in DMSO-*d*₆ (zoomed in)



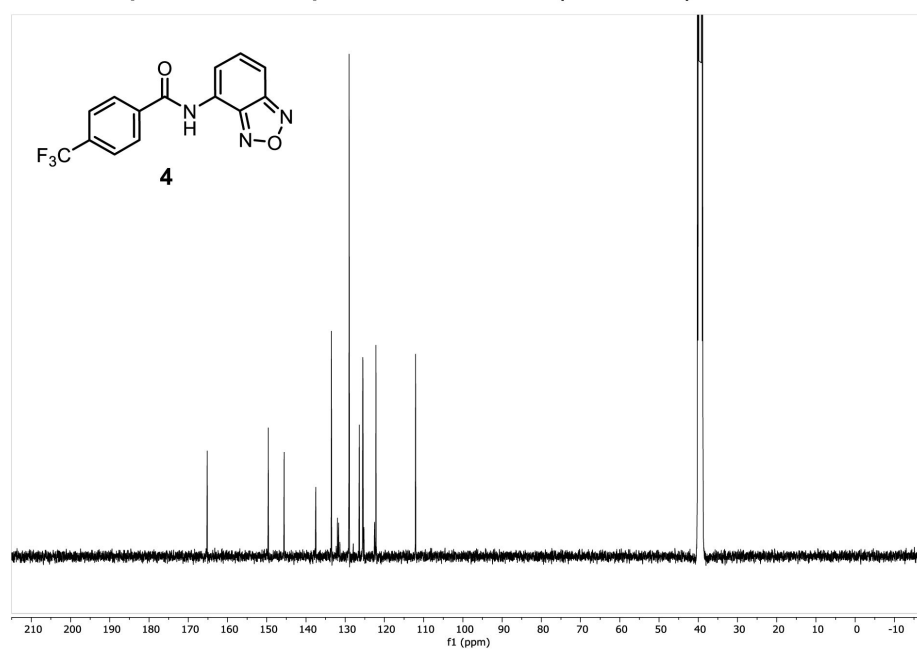
¹H NMR spectrum of compound 4 in DMSO-*d*₆



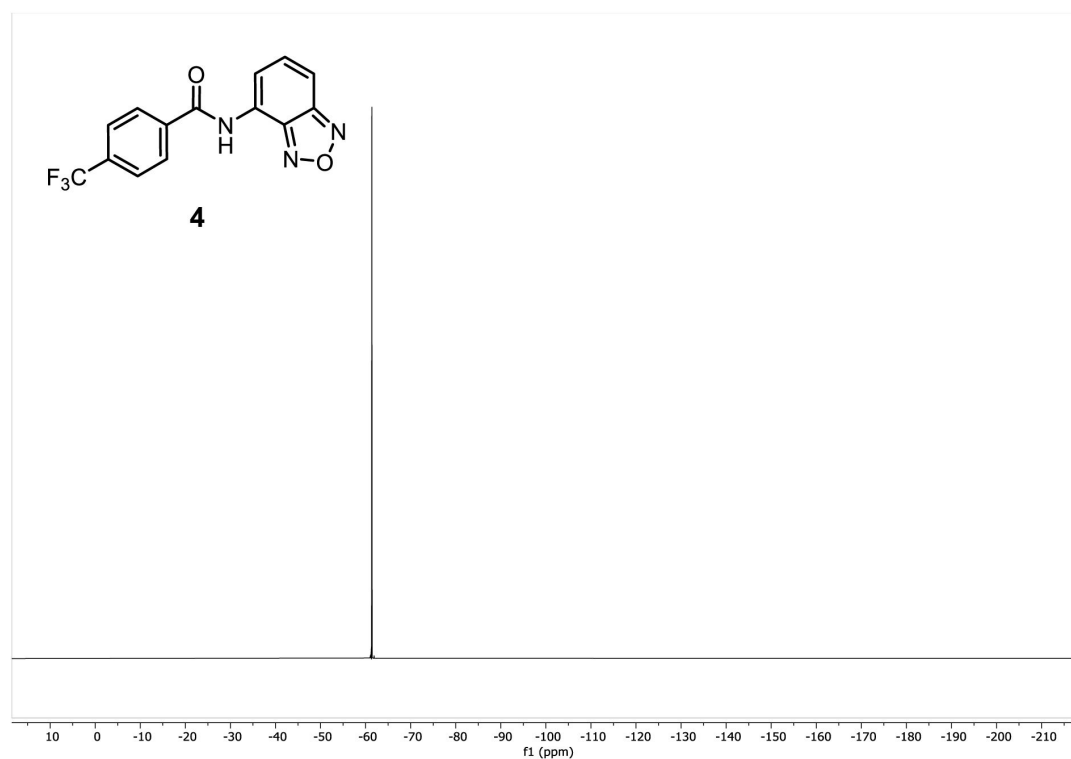
^{13}C NMR spectrum of compound 4 in DMSO-*d*₆



^{13}C NMR spectrum of compound 4 in DMSO-*d*₆ (zoomed in)



¹⁹F NMR spectrum of compound 4 in DMSO-d₆



Datasets

<https://www.rcsb.org/structure/9O48>

<https://www.rcsb.org/structure/9O51>

<https://www.rcsb.org/structure/9O52>

<https://www.rcsb.org/structure/9O53>

<https://www.rcsb.org/structure/9O50>

Supplementary Figures

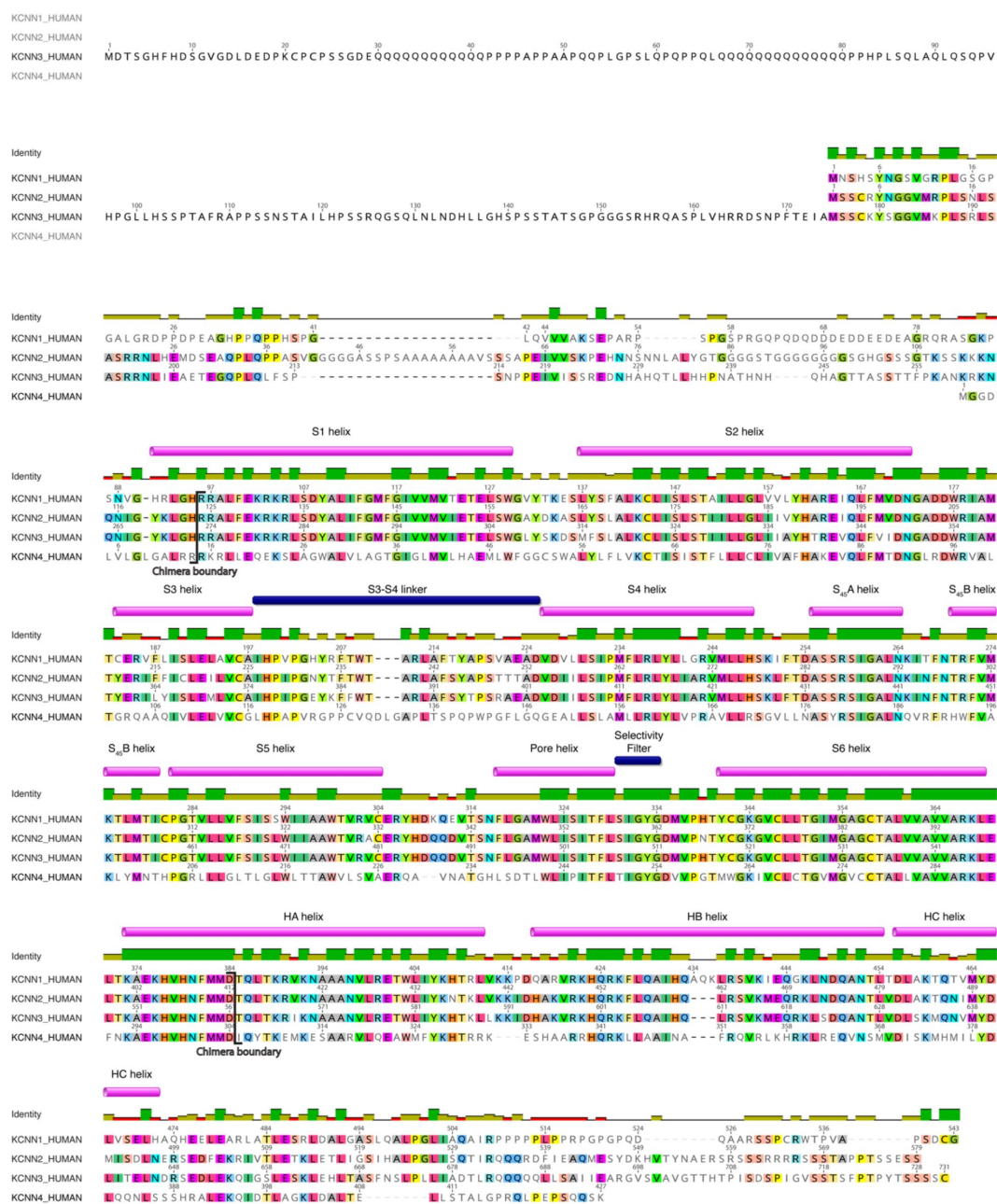


Figure S1

Sequence alignment of human SK1, SK2, SK3, and SK4.

SK2 structural features and chimera boundaries are annotated. Sequence identity is shown in the graph above the sequences.

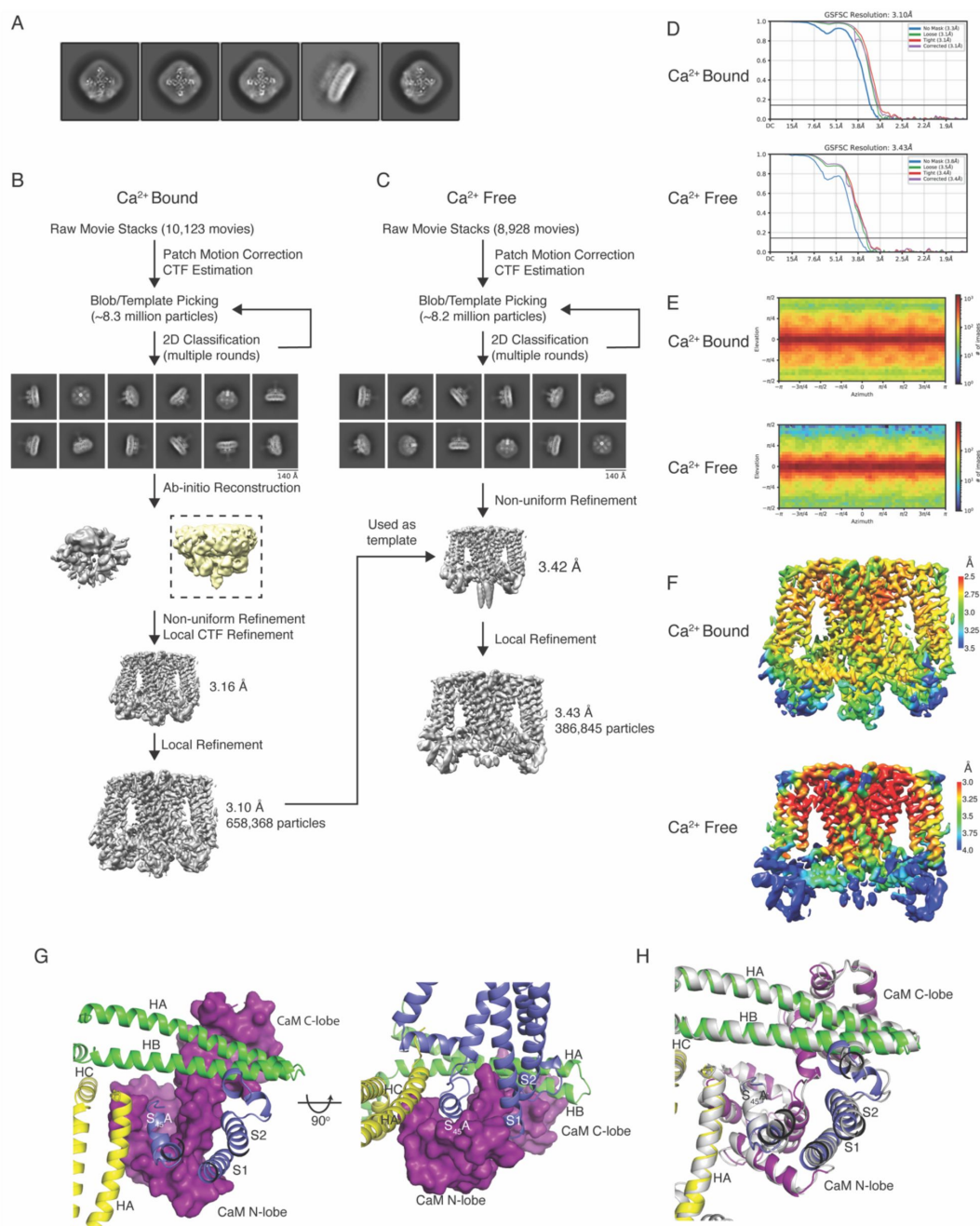


Figure S2

Structure determination of Ca²⁺-bound SK2-4 and Ca²⁺-free SK2-4.

A) 2D class averages of WT SK2 showing disorder in the intracellular region. Cryo-EM structure determination flowchart for (B) Ca²⁺-bound SK2-4 and (C) Ca²⁺-free SK2-4. (D) FSC curves, (E) angular distribution, and (F) local resolution estimates for Ca²⁺-bound SK2-4 and Ca²⁺-free SK2-4. G) Interactions between the SK2-4 intracellular domains and CaM (surface, purple) in the Ca²⁺-bound conformation of SK2-4 (cartoon, each subunit has a different color). The CaM C-lobe interacts with the HA and HB helices and the N-lobe interacts with the S₄₅A helix. H) Overlay of the intracellular domains from SK2-4 (colored as in G) and SK4 (grey) demonstrate a similar CaM binding interaction in each structure.

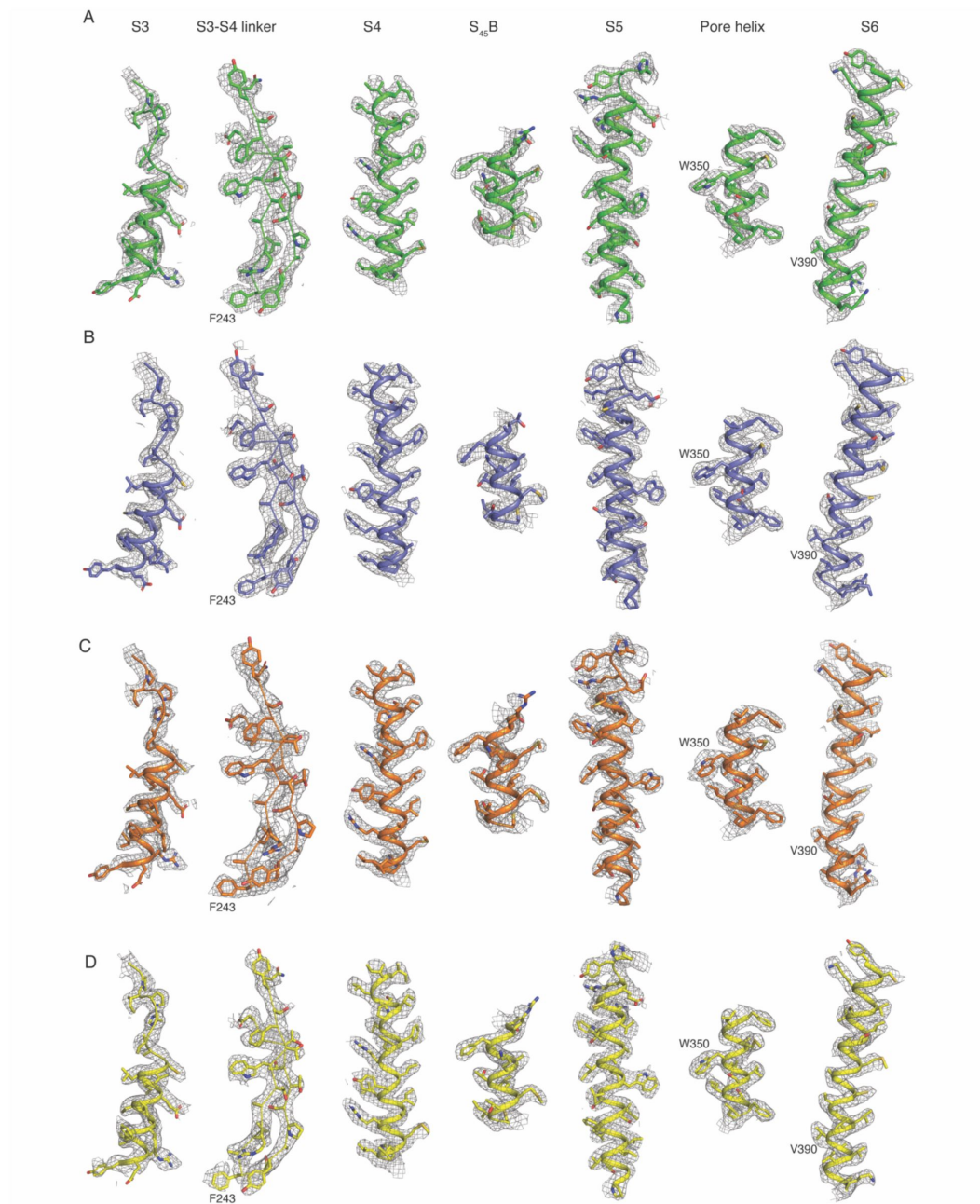


Figure S3

Representative cryo-EM densities.

Cryo-EM density shown at 5 σ of the S3, S3-S4 linker, S4, S₄₅B, S5, pore helix, and S6 for (A) Ca²⁺-bound SK2-4, (B) Ca²⁺-free SK2-4, (C) apamin-bound SK2-4, and (D) compound 1-bound SK2-4.

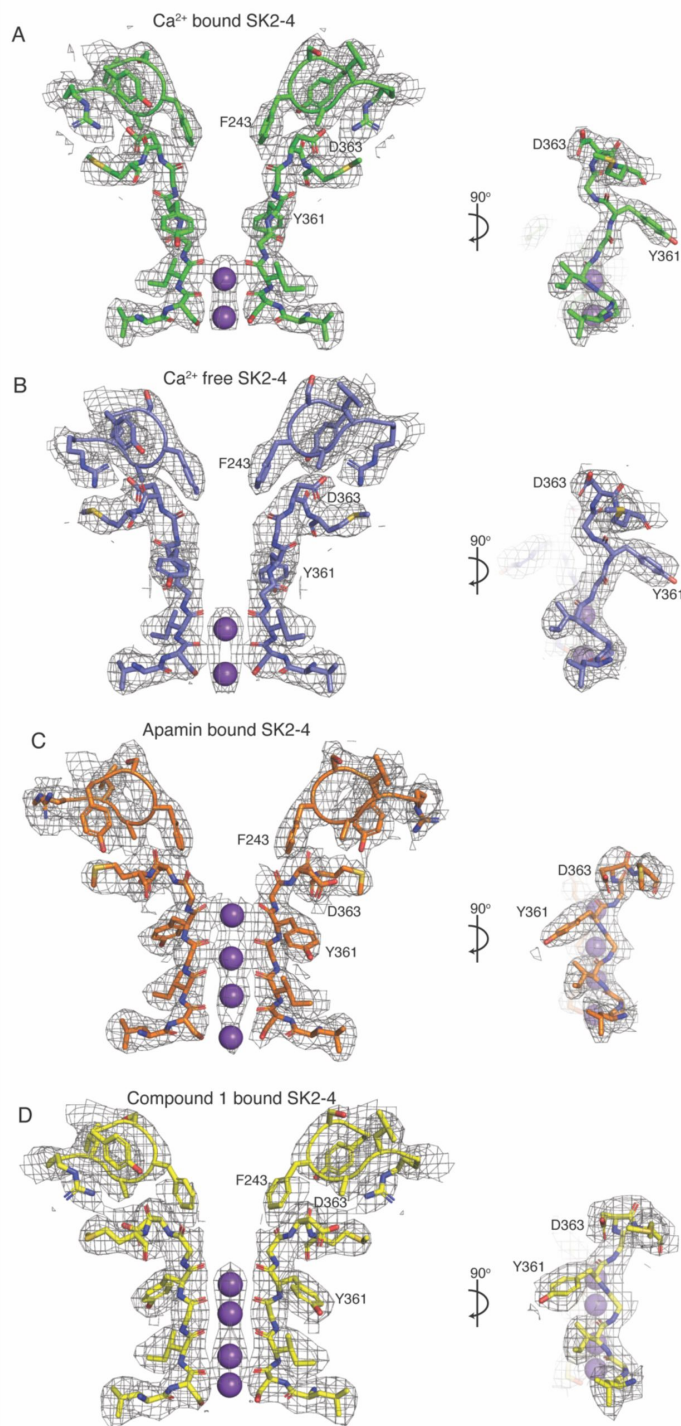


Figure S4

Cryo-EM density for the SK2-4 selectivity filter.

Cryo-EM density shown at 5σ for the selectivity filter and interacting S3-S4 residues (not shown in 90° rotated view) for (A) Ca^{2+} -bound SK2-4, (B) Ca^{2+} -free SK2-4, (C) apamin-bound SK2-4, and (D) compound 1-bound SK2-4.

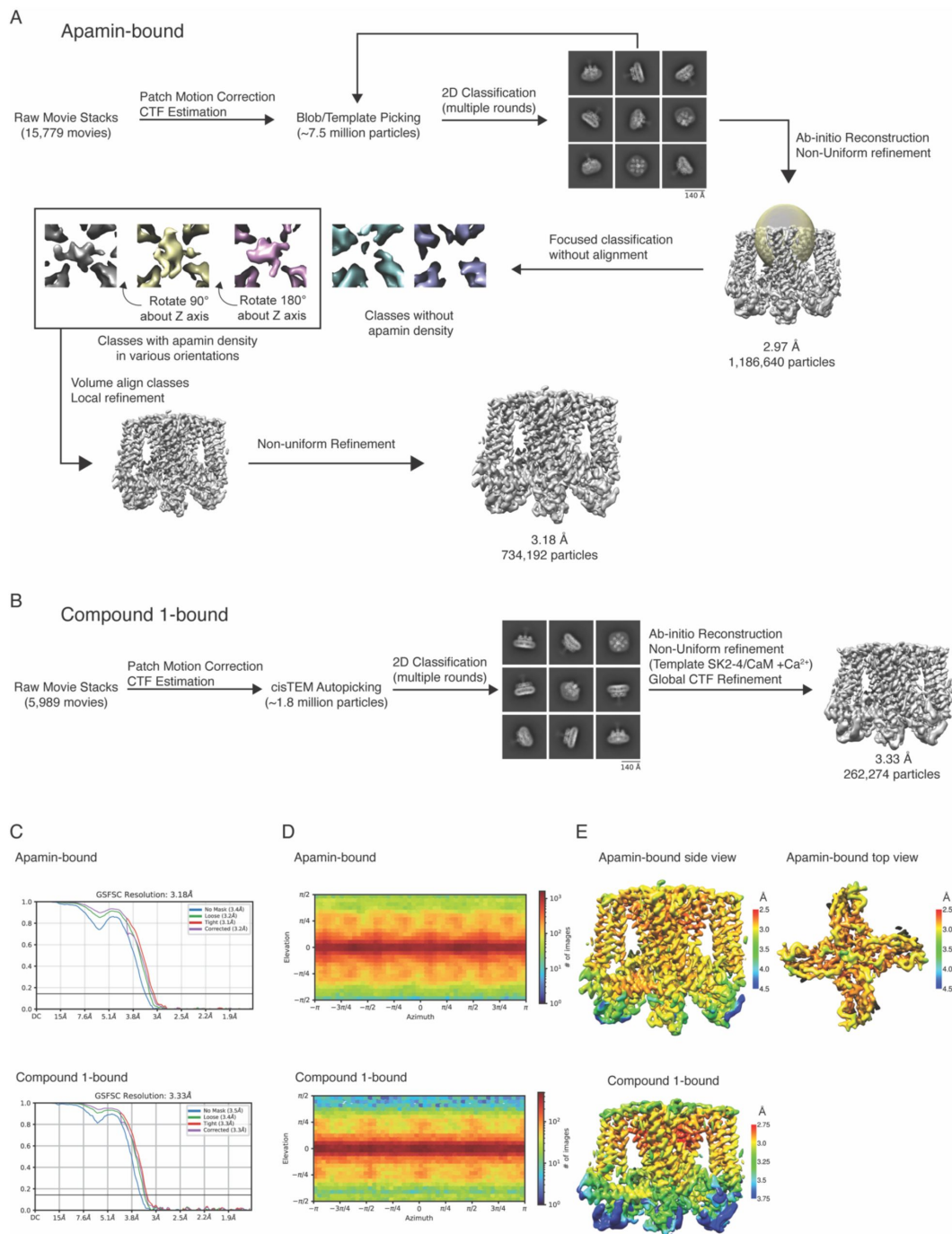


Figure S5

Structure determination of apamin-bound SK2-4 and compound 1-bound SK2-4.

Cryo-EM structure determination flowchart for (A) apamin-bound SK2-4 and (B) compound 1-bound SK2-4. (C) FSC curves, (D) angular distribution, and (E) local resolution estimates for apamin-bound SK2-4 and compound 1-bound SK2-4.

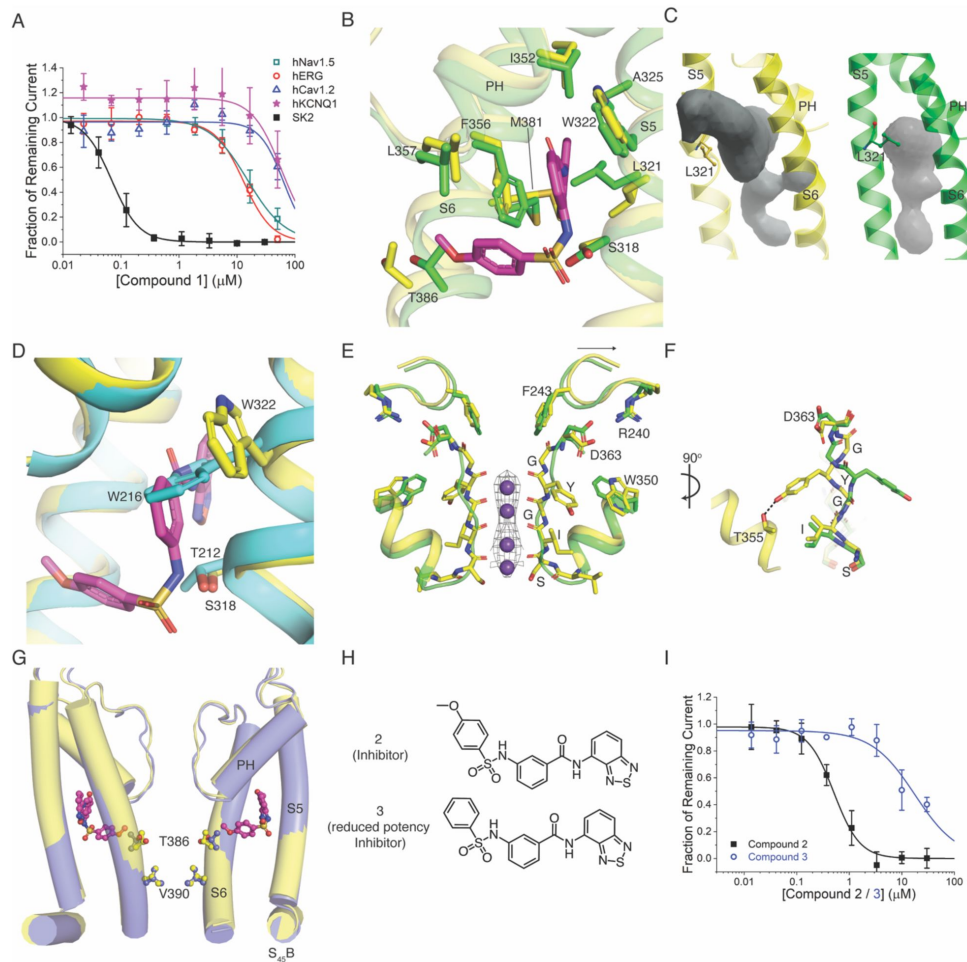


Figure S6

Characterization of compound 1 inhibition.

A) Potency of compound 1 on WT SK2 (black squares, IC_{50} =69 nM), hNav1.5 (cyan squares, IC_{50} =15.3 μM), hERG (red circles, IC_{50} =13 μM), hCav1.2 (blue triangles, IC_{50} >50 μM), and hKCNQ1 channels (purple stars, IC_{50} >50 μM). Data points represent mean $\pm$ SD (n =4-6). See Methods for detailed information regarding ion channel assays. B) Overlay of the compound 1 (magenta sticks) binding site from the structures of compound 1-bound SK2-4 (yellow) and Ca^{2+} -bound SK2-4 (green). A rotation of Leu321 is required to accommodate the benzoxadiazole. The methoxy of compound 1 (magenta sticks) clashes with the S6 Thr386 in Ca^{2+} -bound SK2-4. C) Surface representation of the compound 1 binding pocket in compound 1-bound SK2-4 (left, yellow cartoon and grey pocket) and Ca^{2+} -bound SK2-4 (right, green cartoon and grey pocket). Rotation of Leu321 is required to expand the pocket and accommodate the benzoxadiazole of compound 1. D) Overlay of the compound 1 (magenta sticks) binding site from the structures of compound 1-bound SK2-4 (yellow) and SK4 (cyan, PDB: 6CNM). Two changes in SK4 occlude the compound 1 binding site and may explain the 10-fold lower potency: 1) SK2 Ser318 is replaced by SK4 Thr212 and 2) SK2 Trp322 is conserved but adopts a different rotamer conformation in SK4 (Trp216). E) Overlay of Ca^{2+} -bound SK2-4 selectivity filter (green) and compound 1-bound SK2-4 selectivity filter (yellow, selectivity filter shown as sticks and S3-S4 linker and pore helix shown as cartoons). Upon compound 1 binding the S3-S4 linker moves away from the pore axis (black arrow), Asp363 reorients into hydrogen bonding position with a rotated Trp350 side chain, Tyr361 in the selectivity filter is rotated 180° to form a hydrogen bond with Thr355 in the pore helix, and the selectivity filter adopts a conformation with four K^{+} coordination sites (purple spheres, density shown at 5 σ). F) 90° degree rotation of (E). G) Overlay of Ca^{2+} -free SK2-4 (lavender) and compound 1-bound SK2-4 (yellow). The S6 and S_{4B} helices adopt a similar conformation in the Ca^{2+} -free and compound 1 (magenta spheres)-bound states with a closed intracellular gate (Val390, spheres). H) Structure of compound 2 and compound 3. I) Removal of the methoxy from compound 2 (black squares, IC_{50} =0.51 μM) produces compound 3 (blue circles, IC_{50} =18 μM) with a 30-fold reduction in potency. Data points represent mean $\pm$ SD (n =6). Change in WT hSK2 current amplitude for different cell populations each exposed to a different extracellular compound concentration was normalized to DMSO control to construct dose responses.

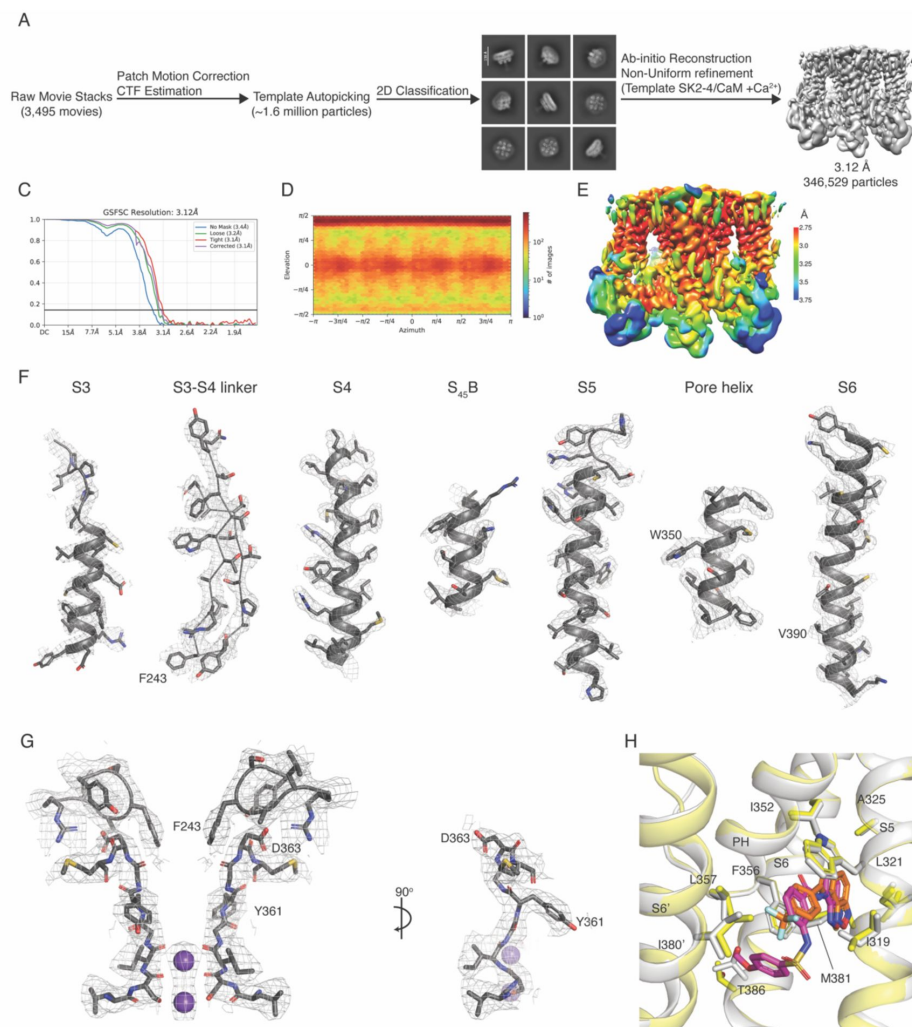


Figure S7

Structure determination of compound 4-bound SK2-4.

Cryo-EM structure determination flowchart (A), FSC curves (B), angular distribution (C), and local resolution estimates (D) for compound 4-bound SK2-4. E) Cryo-EM density shown at 5 σ of the S3, S3-S4 linker, S4, S₄₅B, S5, pore helix, and S6 for compound 4-bound SK2-4. D) Cryo-EM density for the selectivity filter and interacting S3-S4 residues (not shown in 90° rotated view) for compound 4-bound SK2-4. H) Overlay of compound 1 (magenta sticks) bound to SK2-4 (yellow cartoon and sticks) and compound 4 (orange sticks) bound to SK2-4 (grey cartoon and sticks).

Structure	SK2-4/CaM + Ca ²⁺	SK2-4/CaM Ca ²⁺ -free	SK2-4/CaM + apamin	SK2-4/CaM + compound 1	SK2-4/CaM + compound 4
PDB	9O48	9O51	9O52	9O53	9O5O
EMDB	EMD-70089	EMD-70120	EMD-70121	EMD-70122	EMD-70145
Data Collection/Processing					
Microscope	Krios	Krios	Krios	Krios	Glacios
Voltage (kV)	300	300	300	300	200
Magnification	75,000x	75,000x	75,000x	75,000x	120,000x
Nominal defocus range (μM)	-0.8 to -1.6	-0.8 to -1.6	-0.8 to -1.4	-0.8 to -1.6	-0.8 to -2.0
Pixel size (Å)	0.845	0.845	0.845	0.845	0.854
Total dose (e ⁻ /Å ²)	50	50	50	50	35
Exposure time (s)	3	3	3.05	3	28
Micrograph number	10,123	8,928	15,779	7,300	3,495
Symmetry	C4	C4	C1	C4	C4
Initial particle number	~8.3 million	~8.2 million	~7.5 million	~1.9 million	~1.6 million
Final particle number	658,368	386,845	734,192	262,274	346,529
Map resolution (Å)	3.1	3.4	3.2	3.3	3.1
FSC threshold	0.143	0.143	0.143	0.143	0.143
Refinement					
Model Resolution (Å)	3.1	3.5	3.2	3.4	3.2
FSC threshold	0.5	0.5	0.5	0.5	0.5
Map sharpening B-factor (Å ²)	122.9	142.3	126.2	143.6	182.5
Map composition					
Non-hydrogen atoms	16,094	12,302	16,185	15,628	15,742
Protein residues	2,044	1,596	2,056	1,992	1,988
Ligands	0	0	1	4	4
B-factors (Å ²)					
Protein					
SK2-4 chains	46.72	64.36	65.19	30.84	67.53
CaM chains	93.73	145.43	107.20	86.32	134.41
Ligand	N/A	N/A	84.36	11.28	48.74
R.M.S. deviations					
Bond length (Å)	0.002	0.002	0.003	0.004	0.003
Bond angle (°)	0.354	0.400	0.465	0.457	0.5
Validation					
MolProbity score	1.11	1.05	1.06	1.19	1.34
Clashscore	3.19	2.68	2.76	4.11	6.11
Rotamers outliers (%)	0	0	0	0	0
Ramachandran plot (%)					
Favored	98.61	98.71	98.62	98.17	98.83
Allowed	1.39	1.29	1.38	1.83	1.17
Disallowed	0	0	0	0	0

Table S1

Data collection parameters and Refinement statistics

Additional information

Author Contributions

S.J.C, W.L., S.K., and J.R.W. conceived the project and designed the experiments. Cryo-EM data was collected by M.K. and processed by S.J.C, M.K., and J.R.W. Structural analysis was completed by S.J.C, W.A.W, and J.R.W. Electrophysiological characterization of SK2-4, SK2 mutant characterization, HTS screening, and characterization of compounds 1-4 were completed by W.L., Y.T.L., J.H., W.G. Characterization and optimization of compound 1 and compound 4 completed by S.K. and S.P. The manuscript was written by S.J.C, W.L., S.K., and J.R.W. with inputs from all authors.

Note

This reviewed preprint has been updated to correct level headings.

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Reviewer #3 (Public review):

This is a fundamentally important study presenting cryo-EM structures of a human small conductance calcium-activated potassium (SK2) channel in the absence and presence of calcium, or with interesting pharmacological probes bound, including the bee toxin apamin, a small molecule inhibitor, and a small molecule activator. As efforts to solve structures of the wild-type hSK2 channel were unsuccessful, the authors engineered a chimera containing the intracellular domain of the SK4 channel, the subtype of SK channel that was successfully solved in a previous study (reference 13). The authors present many new and exciting findings, including opening of an internal gate (similar to SK4), for the first time resolving the S3-S4 linker sitting atop the outer vestibule of the pore and unanticipated plasticity of the ion selectivity filter, and the binding sites for apamin, one new small molecule inhibitor and another small molecule activator. Appropriate functional data are provided to frame interpretations arising from the structures of the chimeric protein; the data are compelling, the interpretations are sound, and the writing is clear. This high-quality study will be of interest to membrane protein structural biologists, ion channel biophysicists, and chemical biologists, and will be valuable for future drug development targeting SK channels.

Comments on revisions:

The authors have done a nice job of revising the manuscript to address the issues raised in the first round of review and I have no further suggestions.

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Author response:

The following is the authors' response to the original reviews.

Reviewer #1 (Public review):

The small conductance calcium-activated potassium channel 2 (SK2) is an important drug target for treating neurological and cardiovascular diseases. However, structural information on this subtype of SK channels has been lacking, and it has been difficult to draw conclusions about activator and inhibitor binding and action in the absence of structural information.

Here the authors set out to (1) determine the structure of the transmembrane regions of a mammalian SK2 channel, (2) determine the binding site of apamin, a historically important SK2 inhibitor whose mode of action is unclear, and (3) use the structural information to generate a novel set of activators/inhibitors that selectively target SK2.

The authors largely achieved all the proposed goals, and they present their data clearly.

Unable to solve the structure of the human SK2 due to excessive heterogeneity in its cytoplasmic regions, the authors create a chimeric construct using SK4, whose structure was previously solved, and use it for structural studies. The data reveal a unique extracellular structure formed by the S2-S3 loop, which appears to directly interact with the selectivity filter and modulate its conductivity. Structures of SK2 in the absence and presence of the activating Ca^{2+} ions both possess non- K^{+} -selective/conductive selectivity filters, where only sites 3 and 4 are preserved. The S6 gates are captured in closed and open states, respectively. Apamine binds to the S2-S3 loop, and unexpectedly, induces a K^{+} selective/conductive conformation of the selectivity filter while closing the S6 gate.

Through high-throughput screening of small compound libraries and compound optimization, the group identified a reasonably selective inhibitor and a related compound that acts as an activator. The characterization shows that these compounds bind in a novel binding site. Interestingly, the inhibitor, despite binding in a site different from that of apamine, also induces a K⁺ selective/conductive conformation of the selectivity filter while the activator induces a non-K⁺ selective/conductive conformation and an open S6 gate.

The data suggest that the selectivity filter and the S6 gate are rarely open at the same time, and the authors hypothesize that this might be the underlying reason for the small conductance of SK2. The data will be valuable for understanding the mechanism of SK2 channel (and other SK subtypes).

Overall, the data is of good quality and supports the claims made by the authors. However, a deeper analysis of the cryo-EM data sets might yield some important insights, i.e., about the relationship between the conformation of the selectivity filter and the opening of the S6 gate.

We attempted focused 3D classification to identify subsets of particles with the S6 open and the SF in a conductive state but were not able to isolate such a particle class. This indicates that either none or a very small percentage of particles exists in a fully conductive state. This sentence was included in the results section:

“Focused 3D classification of the S3-S4 linker was unsuccessful in identifying particles subsets with a dilated extracellular constriction suggesting that either none or a very small percentage of Ca²⁺-bound SK2-4 is in a conductive state”

Some insight and discussion about the allosteric networks between the SF and the S6 gate would also be a valuable addition.

The extracellular constriction is in the same non-conductive conformation in the Ca²⁺ bound and Ca²⁺-free SK2-4 structures suggesting that the conformation of S3-S4 linker/SF and the S6 are not allosterically coupled. We predict that Ca²⁺ opens the intracellular gate and another physiological factor (not yet identified) promotes extracellular gate opening. These sentences were added to the results and discussion: “This along with the similar conformation of the S3-S4 linker in the Ca²⁺-bound and Ca²⁺-free states of SK2-4 suggest that Ca²⁺-dependent intracellular gate dynamics are not coupled to the conformation of the S3-S4 linker. Other yet to be identified physiological factors may be required to dilate the extracellular constriction.”

“Alternatively, other physiological factors, such as PIP2[46,47] or protein-protein interactions[48-50], may exist in live cells that modulate the interaction between S3-S4 linker and the selectivity filter.”

Reviewer #2 (Public review):

Summary:

The authors have used single-particle cryoEM imaging to determine how small-molecule regulators of the SK channel interact with it and modulate their function.

Strengths:

The reconstructions are of high quality, and the structural details are well described.

Weaknesses:

The electrophysiological data are poorly described. Several details of the structural observations require a mechanistic context, perhaps better relating them to what is known about SK channels or other K channel gating dynamics.

As recommended, additional details for electrophysiological data were added to the results, methods, and figure legends for clarification.

The most pressing point I have to make, which could help improve the manuscript, relates to the selectivity filter (SF) conformation. Whether the two ion-bound state of SK2-4 (Figure 4A) represents a non-selective, conductive SF occluded by F243 or represents a C-type inactivated SF, further occluded by F243, is unclear. It would be important to discuss this. Reconstructions of Kv1.3 channels also feature a similar configuration, which has been correlated to its accelerated C-type inactivation.

Structural overlays of Ca^{2+} bound SK2-4, HCN, and C-type inactivated Kv1.3 selectivity filters demonstrate that each have conformational differences and it is difficult to definitively determine if the SK2-4 selectivity filter is in a non-selective conformation like HCN or a C-type inactivated conformation like Kv1.3. Based on the number of ions observed in the filter and the position of Tyr361 we believe the selectivity filter most closely resembles that of HCN. Importantly, the selectivity filter conformation observed in the SK2-4 Ca^{2+} -bound and Ca^{2+} -free structures is ultimately nonconductive due to the Phe243 extracellular constriction blocking K^+ flux.

A comparison of the SK2-4 selectivity filter to HCN and C-type inactivated Kv1.3 was included in Figure 4 and this sentence was included in the results section:

“The selectivity filter of SK2-4 resembles that of HCN in both the position of Tyr361 and the number of K^+ coordination sites (Fig 4E,F,G,H)”

Furthermore, binding of a toxin derivative to Kv1.3 restores the SF into a conductive form, though occluded by the toxin. It appears that apamin binding to SK2-4 might be doing something similar. Although I am not sure whether SK channels undergo C-type inactivation like gating, classical MTS accessibility studies have suggested that dynamics of the SF might play a role in the gating of SK channels. It would be really useful (if not essential) to discuss the SF dynamics observed in the study and relate them better to aspects of gating reported in the literature.

Extracellular toxin binding to SK2-4 and Kv1.3 induce a conformational change in the selectivity filter to produce a canonical K^+ selective structure with four coordination sites. However, the mechanism by which the toxins produce the conformational change is different. For SK2-4, apamin interacts primarily with S3-S4 linker residues and induces a shift in the S3-S4 linker away from the pore axis. This in turn prevents the hydrogen bonds between Arg240 and Tyr245 of the S3-S4 linker and Asp363 at the C-terminus of the selectivity filter to produce a selectivity filter conformation with four K^+ coordination sites. For Kv1.3, the sea anemone toxin ShK binds directly to the C-terminus of the selectivity filter disrupting interactions required for the C-type inactivated structure and thereby inducing the conformational change. These sentences were added to the results:

“Toxin induced selectivity filter conformational change has also been reported for Kv1.3 with the sea anemone toxin ShK. However, unlike apamin binding to SK2-4, ShK binds directly to the Kv1.3 selectivity filter to convert a C-type inactivated conformation to a canonical K^+ selective structure with four coordination sites [39,40]. The change in selectivity filter conformation in apamin-bound SK2-4 seems to be driven instead by the weakening of interactions between the selectivity filter and the S3-S4 linker.”

The SF of K channels, in conductive states, are usually stabilized by an H-bond network involving water molecules bridged to residues behind the SF (D363 in the down-flipped conformation and Y361). Considering the high quality of the reconstructions, I would suspect that the authors might observe speckles of density (possibly in their sharpened map) at these sites, which overlap with water molecules identified in high-resolution X-ray structures of KcsA, MthK, NaK, NaK2K, etc. It could be useful to inspect this region of the density map.

We did not observe strong density near Y361 or D363 that could be confidently model as water. However, in the structures of SK2-4 bound to apamin and compound 1 Tyr361 in the selectivity filter rotates 180° and forms a hydrogen bond with Thr355 in the pore helix. The homologous hydrogen bond is also observed in SK4 and the conductive/ K⁺ selective selectivity filter conformation of Kv1.3. The rotation of Tyr361 to form a hydrogen bond with Thr355, reorientation of Asp363 and Trp350 into hydrogen bonding position, and the presence of four K⁺ coordination sites upon binding of apamin and compound 1 strongly suggest that the selectivity filter is in a K⁺ selective/conductive conformation. The Tyr361/Thr355 hydrogen bond is now described in the paper and shown in Figures 4D, 5D, and S6F.

Reviewer #3 (Public review):

This is a fundamentally important study presenting cryo-EM structures of a human small conductance calcium-activated potassium (SK2) channel in the absence and presence of calcium, or with interesting pharmacological probes bound, including the bee toxin apamin, a small molecule inhibitor, and a small molecule activator. As eOorts to solve structures of the wild-type hSK2 channel were unsuccessful, the authors engineered a chimera containing the intracellular domain of the SK4 channel, the subtype of SK channel that was successfully solved in a previous study (reference 13). The authors present many new and exciting findings, including opening of an internal gate (similar to SK4), for the first time resolving the S3-S4 linker sitting atop the outer vestibule of the pore and unanticipated plasticity of the ion selectivity filter, and the binding sites for apamin, one new small molecule inhibitor and another small molecule activator. Appropriate functional data are provided to frame interpretations arising from the structures of the chimeric protein; the data are compelling, the interpretations are sound, and the writing is clear. This high-quality study will be of interest to membrane protein structural biologists, ion channel biophysicists, and chemical biologists, and will be valuable for future drug development targeting SK channels.

The following are suggestions for strengthening an already very strong and solid manuscript:

(1) It would be good to include some information in the text of the results section about the method and configuration used to obtain electrophysiological data and the limitations. It is not until later in the text that the Qube instrument is mentioned in the results section, and it is not until the methods section that the reader learns it was used to obtain all the electrophysiological data. Even there, it is not explicitly mentioned that a series of diOerent internal solutions were used in each cell where the free calcium concentration was varied to obtain the data in Figure1C. Also, please state the concentration of free calcium for the data in Figure 1B.

As recommended, additional details for electrophysiological data were added to the results, methods, and figure legends for clarification.

(2) The authors do a nice job of discussing the conformations of the selectivity filter they observed here in SK as they relate to previous work on NaK and HCN, but from my perspective the authors are missing an opportunity to point out even more striking relationships with slow C-type inactivation of the selectivity filter in Shaker and Kv1 channels. C-type inactivation of the filter in Shaker was seen in 150 mM K using the W434F mutant (PMC8932672) or in 4 mM K for the WT channel (PMC8932672), and similar results have been reported for Kv1.2 (PMC9032944; PMC11825129) and for Kv1.3 (PMC9253088; PMC8812516) channels. For Kv1.3, C-type inactivation occurs even in 150 mM K (PMC9253088; PMC8812516). Not unlike what is seen here with apamin, binding of the sea anemone toxin (ShK) with a Fab attached (or the related dalazatide) inserts a Lys into the selectivity filter and stabilizes the conducting conformation of Kv1.3 even though the Lys depletes occupancy of S1 by potassium (PMC9253088; PMC8812516). Or might the conformation of the filter be controlled by regulatory processes in SK2 channels? I think connecting the dots here would enhance the impact of this study, even if it remains relatively speculative.

Please see the response to reviewer 2's comments for a comparison of the selectivity filter structure between SK2-4 and C-type inactivated Kv1.3 and a discussion of toxin induced selectivity filter conformational change.

What is known about how the functional properties of SK2 channels (where the filter changes conformation) differ from SK4, where the filter remains conducting (reference 13)? Is there any evidence that SK2 channels inactivate?

Compared with SK4, SK2 has some unique properties such as lower conductance and the ability to switch between low- and high-open probability states. Mutation of Phe243 suggests that the S3-S4 linker conformation contributes to the low conductance. This is included in the discussion.

“Such a mechanism may explain some properties of SK2 that are not observed in SK4, which lacks an S3-S4 linker, such as its low conductance (~10 pS) and the ability to switch between low- and high-open probability states[3,4]. Indeed, mutation of Phe243 in rat SK2 produced a 2-fold increase in channel conductance[5].”

Or might the conformation of the filter be controlled by regulatory processes in SK2 channels? I think connecting the dots here would enhance the impact of this study, even if it remains relatively speculative.

Please see the response to reviewer 1's comments for a discussion of the potential physiological role of the S3-S4 linker/extracellular constriction and its mechanism for opening.

Reviewer #1 (Recommendations for the authors):

I enjoyed reading your paper and am intrigued by your findings on the selectivity filter of SK2. I've got a few recommendations for data analysis and a couple of questions that might contribute to the discussion.

In your Ca²⁺-bound dataset, have you tried to parse out any alternative conformations (e.g., by using 3D classification, or 3D variability)? Do you think there might be a small(er) population of particles that adopt a fully open conformation? If you haven't done this already, I would recommend doing so. You have a rather large number of particles in your final 3D reconstruction (~660k), so there might be some hidden conformations that could contribute to our understanding of the system.

I would recommend doing the same for your compound 4-bound data set.

Please see above for response to this recommendation.

Do you think apamine works solely as a pore blocker, or does its binding perhaps also affect the S6 gate via allosteric networks (perhaps the same ones that induce the formation of the K⁺ conductive SF through binding of compound 1 above the S6 gate?)?

Apamin binding does not change the conformation of the pore helices (S5 or S6) and thus we believe it acts primarily as a pore blocker. The following was added to the results section:

“Overall, the apamin-bound SK2-4/CaM structure resembles Ca²⁺-bound SK2-4. The Nterminal lobe of CaM engages with the S₄₅ A helix, the S5 and S6 helices adopt a similar conformation, and the intracellular gate Val390 is open with a radius of 3.5 Å (Fig 2D). The most significant conformational change is in the position of the S3-S4 linker, which shifts ~2 Å away from the pore axis to accommodate apamin binding.”

Is there a mechanistic explanation for why it might be difficult/energetically costly for the SF to be conductive and the S6 gate to be open at the same time?

Not to our knowledge.

I also have these minor recommendations:

-In all figures showing density, include the threshold/sigma value at which density is shown.

-For all ligands and ions, include half-map data.

Sigma values were added for all figures legends displaying cryoEM density. The displayed maps are the sharpened full maps.

Reviewer #2 (Recommendations for the authors):

Is it possible to provide a structure-sequence guided explanation for the different affinity of compound 1 for SK2 vs SK4?

Yes. The following is now included in the results section and a panel was added to Figure S6D.

“However, for SK4 Thr212 replaces SK2 Ser318 and Trp216 (homologous to SK2 Trp322) is conserved but adopts a different rotamer conformation (Fig S6D). Both changes occlude the compound 1 binding site in SK4 and would likely reduce compound 1 potency on SK4 as observed in the functional data.”

Is it possible to propose a model of modulation by compound 1/4 where the authors can comment on the conformational dependence of compound binding? That is, do they bind exclusively to the identified conformational states of the channel, or are they able to bind to both closed and open channels, but bias one state over the other?

The clash between compound 1 and Thr386 in the open conformation of the S6 helices suggests that compound 1 would preferentially bind to closed state of SK2. Similarly, the clash between compound 4 and Ile380 in the closed conformation of the S6 helices suggests that compound 4 would preferentially bind to the open state of SK2. This was included in the discussion:

“This proposed mechanism of modulation suggests that compound 1 may bind preferentially to the closed conformation of the S6 helices and compound 4 may bind preferentially to the open conformation of the S6 helices.”

Please provide the calcium concentration used to generate the data in Figure 1B. The calcium concentration is now stated in the legend for Fig 1B:

“Intracellular solution contains 2 μM Ca^{2+} based on calculation using Maxchelator (see methods)”

Essential and critically important descriptions of experiments in Figure 7A are lacking. It would be essential to describe properly, with care, what the currents and the conditions of measurements are. If these currents are obtained by subtracting leak currents by adding other drugs, it would be good to comment on whether the latter compete with compounds 1/4.

As recommended, additional details for electrophysiological data were added to the results, methods, and figure legends for clarification. SK currents were obtained by subtracting leak currents by adding UCL1684 only at the end of experiments. UCL1684 is not expected to interfere with eVect of compound 1 or 4 given diVerent binding sites and mechanisms.

If Compound 1 changes the structure of the SF (Figure 6F), would it also promote apamin binding? Given that both these agents produce a similar change in the SF, could each favor the binding of the other?

Since apamin binds to the S3-S4 linker it is unlikely that the selectivity filter conformational change observed in the compound 1 bound structure would aVect apamin binding.

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